

V. CONCLUSIONS AND SUGGESTIONS

1. Conclusions

Durian shell was found as a potential sorbent to remove Chromium (VI) from contaminated solution. The adsorption capacity of durian shell to Chromium (VI) represented by the maximum adsorption capacity is approximately about 16 mg/g at low pH and about 6 mg/g at higher pH.

Both Langmuir and Freundlich models represent the adsorption equilibria data quite-well. While for the kinetic studies, pseudo-second order kinetic model gave better correlation for kinetic data in comparison to pseudo first-order model. From thermodynamic studies, were found that the sorption process run spontaneously, endothermic and irreversible.

The biosorption process of Chromium (VI) on durian shell is dominated by chemical sorption.

2. Suggestions

The studies of biosorption process of Chromium (VI) using durian shell need further experiments like chemical treatment on durian shell to improve the adsorption capacity of durian shell and to regenerate the durian shell after used as the adsorbent, so it is more economical to use in wastewater treatment.

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