

BAB V **KESIMPULAN DAN SARAN**

V.1 Kesimpulan

Berdasarkan penelitian yang dilakukan, diperoleh komposit sel/Zn-BDC berbasis selulosa dari kulit durian didapatkan kesimpulan:

1. Selulosa berbasis kulit durian hasil pretreatment memiliki kemurnian sebesar 77,9% dan dapat digunakan untuk membentuk Sel/Zn-BDC. Komposit ini memiliki diameter dan panjang 200 nm dan 400 nm, serta memiliki bentuk seperti batang (*rod*). Sel/Zn-BDC memiliki luas area spesifik sebesar $12,24 \text{ m}^2/\text{g}$. Karakteristik ini didapatkan dengan perbandingan rasio selulosa: logam: ligand sebesar 2: 11: 2,5.
2. Studi kinetika yang sesuai untuk adsorpsi ini adalah *pseudo-second order* yang menandakan adsorpsi yang berlangsung dikendalikan oleh adsorpsi kimia. Adsorpsi dilakukan pada kondisi suhu 30°C , pH 3 dan selama 8 jam. Dari percobaan didapatkan R^2 sebesar 0,9894 pada suhu 30°C dan kapasitas maksimum *tetracycline* yang dapat diadsorpsi oleh sel/Zn-BDC sebesar $1173,6 \text{ mg. g}^{-1}$.
3. Studi isotherm adsorpsi yang menggunakan persamaan Freundlich menandakan bahwa permukaan dari sel/Zn-BDC tidak heterogen dan sistem adsorpsi *unfavorable*. Sementara itu, analisa BET didapatkan R^2 sebesar 0,9948 pada suhu 30°C menandakan bahwa mekanisme adsorpsinya adalah multilayer. Jumlah lapisan yang didapatkan dari persamaan BET ini adalah 20,99 lapisan (30°C); 24,93 lapisan (40°C); 41,97 lapisan (50°C). Dari analisa Harkin-Jura

diketahui bahwa orientasi tumbukan adsorpsi *tetracycline* dan sel/Zn-BDC terjadi secara vertikal.

4. Berdasarkan hasil perhitungan studi termodinamika didapatkan bahwa adsorpsi bersifat eksotermis. Adsorpsi ini berlangsung secara *chemisorption* yang dimulai secara spontan pada layer pertama dan menjadi tidak spontan di layer teratas.

V.2 Saran

Pada proses pretreatment kulit durian, khususnya setelah proses delignifikasi, dibutuhkan pencucian hingga pH kulit durian netral. Prosedur pencucian dapat dilakukan melalui siklus berulang perendaman dan penggantian air baru hingga pH mencapai netral. Hal ini agar dapat mengurangi kebutuhan akuades dan tenaga yang digunakan. Pada proses sintesa, pengadukan selulosa, logam dan ligand dapat dilakukan bersamaan dengan proses sonikasi untuk memastikan partikel MOF dapat terdistribusi dengan baik pada permukaan selulosa.

DAFTAR PUSTAKA

- [1] M. Stylianou, A. Christou, C. Michael, A. Agapiou, P. Papanastasiou, and D. Fatta-Kassinos, “Adsorption and removal of seven antibiotic compounds present in water with the use of biochar derived from the pyrolysis of organic waste feedstocks,” *J. Environ. Chem. Eng.*, vol. 9, no. 5, p. 105868, 2021, doi: 10.1016/j.jece.2021.105868.
- [2] T. P. Van Boeckel *et al.*, “Global trends in antimicrobial use in food animals,” *Proc. Natl. Acad. Sci. U. S. A.*, vol. 112, no. 18, pp. 5649–5654, 2015, doi: 10.1073/pnas.1503141112.
- [3] R. Daghrir and P. Drogui, “Tetracycline antibiotics in the environment: A review,” *Environ. Chem. Lett.*, vol. 11, no. 3, pp. 209–227, 2013, doi: 10.1007/s10311-013-0404-8.
- [4] Y. L. Zhang, S. S. Lin, C. M. Dai, L. Shi, and X. F. Zhou, “Sorption-desorption and transport of trimethoprim and sulfonamide antibiotics in agricultural soil: Effect of soil type, dissolved organic matter, and pH,” *Environ. Sci. Pollut. Res.*, vol. 21, no. 9, pp. 5827–5835, 2014, doi: 10.1007/s11356-014-2493-8.
- [5] S. Hokkanen, A. Bhatnagar, and M. Sillanpää, “A review on modification methods to cellulose-based adsorbents to improve adsorption capacity,” *Water Res.*, vol. 91, pp. 156–173, 2016, doi: 10.1016/j.watres.2016.01.008.
- [6] P.-S. Keng, S.-L. Lee, S.-T. Ha, Y.-T. Hung, and S.-T. Ong, *Cheap Materials to Clean Heavy Metal Polluted Waters*. 2013. doi: 10.1007/978-94-007-6836-9_8.
- [7] D. J. Gardner, G. S. Oporto, R. Mills, and M. A. S. A. Samir, “Adhesion and surface issues in cellulose and nanocellulose,” *J. Adhes. Sci. Technol.*, vol. 22, no. 5–6, pp. 545–567, 2008, doi: 10.1163/156856108X295509.
- [8] N. Amiralian *et al.*, “Magnetic nanocellulose: A potential material for removal of dye from water,” *J. Hazard. Mater.*, vol. 394, no. February, p. 122571, 2020, doi: 10.1016/j.jhazmat.2020.122571.
- [9] P. Penjumras, R. B. A. Rahman, R. A. Talib, and K. Abdan, “Extraction and Characterization of Cellulose from Durian Rind,” *Agric. Agric. Sci. Procedia*, vol. 2, pp. 237–243, 2014, doi:

- 10.1016/j.aaspro.2014.11.034.
- [10] S. K. Karmelia Nurrohmah¹, Arni Komala Sari¹, Dina Riziani¹, “MAKUDU (Makaroni Kulit Durian): POTENSI PANGAN OLAHAN PRAKTIS UNTUK MENGURANGI LIMBAH KULIT DURIAN,” *Jitipari*, vol. 6, no. 1, pp. 30–40, 2021.
- [11] Z. Guo, F. Yang, R. Yang, L. Sun, Y. Li, and J. Xu, “Preparation of novel ZnO-NP@Zn-MOF-74 composites for simultaneous removal of copper and tetracycline from aqueous solution,” *Sep. Purif. Technol.*, vol. 274, no. April, p. 118949, 2021, doi: 10.1016/j.seppur.2021.118949.
- [12] Farida Hanum, Nurhasmawaty Pohan, Mulia Rambe, Ratih Primadony, and Mei Ulyana, “Pengaruh Massa Ragi Dan Waktu Fermentasi Terhadap Bioetanol Dari Biji Durian,” *J. Tek. Kim. USU*, vol. 2, no. 4, pp. 49–54, 2013, doi: 10.32734/jtk.v2i4.1491.
- [13] R. D. Kusumaningtyas and A. F. A. Syah, “Conversion of Durian Shell Agroindustrial Waste Into Various Valuable Products To Support the Food Security During the Covid-19 New Normal Era: Review,” *J. Teknol. Has. Pertan.*, vol. 13, no. 2, p. 111, 2020, doi: 10.20961/jthp.v13i2.43599.
- [14] A. K. Obeng, D. Premjet, and S. Premjet, “Fermentable sugar production from the peels of two durian (*Durio zibethinus* Murr.) cultivars by phosphoric acid pretreatment,” *Resources*, vol. 7, no. 4, 2018, doi: 10.3390/resources7040060.
- [15] Z. M. Lazim, T. Hadibarata, M. H. Puteh, Z. Yusop, R. Wirasnita, and N. Mohd Nor, “Utilization of durian peel as potential adsorbent for bisphenol a removal in aquoeus solution,” *J. Teknol.*, vol. 74, no. 11, pp. 109–115, 2015, doi: 10.11113/jt.v74.4879.
- [16] I. C. Yadav and N. L. Devi, “Biomass burning, regional air quality, and climate change,” *Encycl. Environ. Heal.*, no. May, pp. 386–391, 2019, doi: 10.1016/B978-0-12-409548-9.11022-X.
- [17] H. M. Jang and E. Kan, “Engineered biochar from agricultural waste for removal of tetracycline in water,” *Bioresour. Technol.*, vol. 284, no. February, pp. 437–447, 2019, doi: 10.1016/j.biortech.2019.03.131.

- [18] R. Rusli and S. J. Eichhorn, “Determination of the stiffness of cellulose nanowhiskers and the fiber-matrix interface in a nanocomposite using Raman spectroscopy,” *Appl. Phys. Lett.*, vol. 93, no. 3, 2008, doi: 10.1063/1.2963491.
- [19] K. Y. Foo and B. H. Hameed, “An overview of dye removal via activated carbon adsorption process,” *Desalin. Water Treat.*, vol. 19, no. 1–3, pp. 255–274, 2010, doi: 10.5004/dwt.2010.1214.
- [20] V. Gómez, M. S. Larrechi, and M. P. Callao, “Kinetic and adsorption study of acid dye removal using activated carbon,” *Chemosphere*, vol. 69, no. 7, pp. 1151–1158, 2007, doi: 10.1016/j.chemosphere.2007.03.076.
- [21] W. Gu *et al.*, “High-efficiency adsorption of tetracycline by cooperation of carbon and iron in a magnetic Fe/porous carbon hybrid with effective Fenton regeneration,” *Appl. Surf. Sci.*, vol. 538, no. August 2020, p. 147813, 2021, doi: 10.1016/j.apsusc.2020.147813.
- [22] J. Sun, L. Cui, Y. Gao, Y. He, H. Liu, and Z. Huang, “Environmental application of magnetic cellulose derived from *Pennisetum sinense Roxb* for efficient tetracycline removal,” *Carbohydr. Polym.*, vol. 251, no. May 2020, p. 117004, 2021, doi: 10.1016/j.carbpol.2020.117004.
- [23] A. Liu, J. Liu, S. He, J. Zhang, and W. Shao, “International Journal of Biological Macromolecules Bimetallic MOFs loaded cellulose as an environment friendly bioadsorbent for highly efficient tetracycline removal,” vol. 225, no. September 2022, pp. 40–50, 2023.
- [24] M. Börjesson and G. Westman, “Crystalline Nanocellulose — Preparation, Modification, and Properties,” *Cellul. - Fundam. Asp. Curr. Trends*, 2015, doi: 10.5772/61899.
- [25] B. Yuan *et al.*, “Nanocellulose-based composite materials for wastewater treatment and waste-oil remediation,” *ES Food Agrofor.*, pp. 41–52, 2020, doi: 10.30919/esfaf0004.
- [26] A. Qiao *et al.*, “Advances in nanocellulose-based materials as adsorbents of heavy metals and dyes,” *Carbohydr. Polym.*, vol. 272, no. July, 2021, doi: 10.1016/j.carbpol.2021.118471.

- [27] M. E. Pasaoglu and I. Koyuncu, “Substitution of petroleum-based polymeric materials used in the electrospinning process with nanocellulose: A review and future outlook,” *Chemosphere*, vol. 269, 2021, doi: 10.1016/j.chemosphere.2020.128710.
- [28] D. Trache *et al.*, “Microcrystalline cellulose: Isolation, characterization and bio-composites application—A review,” *Int. J. Biol. Macromol.*, vol. 93, pp. 789–804, 2016, doi: 10.1016/j.ijbiomac.2016.09.056.
- [29] Z. Jiang, S. H. Ho, X. Wang, Y. Li, and C. Wang, “Application of biodegradable cellulose-based biomass materials in wastewater treatment,” *Environ. Pollut.*, vol. 290, no. September, p. 118087, 2021, doi: 10.1016/j.envpol.2021.118087.
- [30] X. Z. Guo, S. S. Han, J. M. Yang, X. M. Wang, S. S. Chen, and S. Quan, “Effect of Synergistic Interplay between Surface Charge, Crystalline Defects, and Pore Volume of MIL-100(Fe) on Adsorption of Aqueous Organic Dyes,” *Ind. Eng. Chem. Res.*, vol. 59, no. 5, pp. 2113–2122, 2020, doi: 10.1021/acs.iecr.9b05715.
- [31] H. Lu, L. Zhang, J. Ma, N. Alam, X. Zhou, and Y. Ni, “Nano-cellulose/mof derived carbon doped cuo/fe₃o₄ nanocomposite as high efficient catalyst for organic pollutant remedy,” *Nanomaterials*, vol. 9, no. 2, 2019, doi: 10.3390/nano9020277.
- [32] J. Zhang and K. Liu, “Enhanced Adsorption of Doxycycline Hydrochloride (DCH) from Water on Zeolitic Imidazolate Framework-8 Modified by Cu²⁺ (Cu-ZIF-8),” *Water. Air. Soil Pollut.*, vol. 231, no. 2, 2020, doi: 10.1007/s11270-020-4455-8.
- [33] R. Zhao, T. Ma, S. Zhao, H. Rong, Y. Tian, and G. Zhu, “Uniform and stable immobilization of metal-organic frameworks into chitosan matrix for enhanced tetracycline removal from water,” *Chem. Eng. J.*, vol. 382, p. 122893, 2020, doi: 10.1016/j.cej.2019.122893.
- [34] X. Zhao, M. Zheng, X. Gao, J. Zhang, E. Wang, and Z. Gao, “The application of MOFs-based materials for antibacterials adsorption,” *Coord. Chem. Rev.*, vol. 440, p. 213970, 2021, doi: 10.1016/j.ccr.2021.213970.
- [35] J. Jin *et al.*, “Cu and Co nanoparticles co-doped MIL-101 as a novel

- adsorbent for efficient removal of tetracycline from aqueous solutions,” *Sci. Total Environ.*, vol. 650, pp. 408–418, 2019, doi: 10.1016/j.scitotenv.2018.08.434.
- [36] J. Xia, Y. Gao, and G. Yu, “Tetracycline removal from aqueous solution using zirconium-based metal-organic frameworks (Zr-MOFs) with different pore size and topology: Adsorption isotherm, kinetic and mechanism studies,” *J. Colloid Interface Sci.*, vol. 590, pp. 495–505, 2021, doi: 10.1016/j.jcis.2021.01.046.
- [37] Y. Zhang, J. Zhou, J. Chen, X. Feng, and W. Cai, “Rapid degradation of tetracycline hydrochloride by heterogeneous photocatalysis coupling persulfate oxidation with MIL-53(Fe) under visible light irradiation,” *J. Hazard. Mater.*, vol. 392, no. December 2019, p. 122315, 2020, doi: 10.1016/j.jhazmat.2020.122315.
- [38] S. Fujisawa, T. Ikeuchi, M. Takeuchi, T. Saito, and A. Isogai, “Superior reinforcement effect of TEMPO-oxidized cellulose nanofibrils in polystyrene matrix: optical, thermal, and mechanical studies.,” *Biomacromolecules*, vol. 13, no. 7, pp. 2188–2194, 2012, doi: 10.1021/bm300609c.
- [39] Q. Ren, F. Wei, H. Chen, D. Chen, and B. Ding, “Preparation of Zn-MOFs by microwave-assisted ball milling for removal of tetracycline hydrochloride and Congo red from wastewater,” *Green Process. Synth.*, vol. 10, no. 1, pp. 125–133, 2021, doi: 10.1515/gps-2021-0020.
- [40] D. J. Tranchemontagne, J. R. Hunt, and O. M. Yaghi, “Room temperature synthesis of metal-organic frameworks: MOF-5, MOF-74, MOF-177, MOF-199, and IRMOF-0,” *Tetrahedron*, vol. 64, no. 36, pp. 8553–8557, 2008, doi: 10.1016/j.tet.2008.06.036.
- [41] S. Khanjani and A. Morsali, “Ultrasound-promoted coating of MOF-5 on silk fiber and study of adsorptive removal and recovery of hazardous anionic dye ‘congo red,’” *Ultrason. Sonochem.*, vol. 21, no. 4, pp. 1424–1429, 2014, doi: 10.1016/j.ultsonch.2013.12.012.
- [42] N. Tian *et al.*, “The synthesis of mesostructured NH₂-MIL-101(Cr) and kinetic and thermodynamic study in tetracycline aqueous solutions,” *J. Porous Mater.*, vol. 23, no. 5, pp. 1269–1278, 2016, doi: 10.1007/s10934-016-0186-z.

- [43] C. S. Wu, Z. H. Xiong, C. Li, and J. M. Zhang, “Zeolitic imidazolate metal organic framework ZIF-8 with ultra-high adsorption capacity bound tetracycline in aqueous solution,” *RSC Adv.*, vol. 5, no. 100, pp. 82127–82137, 2015, doi: 10.1039/c5ra15497a.
- [44] S. Álvarez-Torrellas, R. S. Ribeiro, H. T. Gomes, G. Ovejero, and J. García, “Removal of antibiotic compounds by adsorption using glycerol-based carbon materials,” *Chem. Eng. J.*, vol. 296, pp. 277–288, 2016, doi: 10.1016/j.cej.2016.03.112.
- [45] C. Chen, D. Chen, S. Xie, H. Quan, X. Luo, and L. Guo, “Adsorption Behaviors of Organic Micropollutants on Zirconium Metal-Organic Framework UiO-66: Analysis of Surface Interactions,” *ACS Appl. Mater. Interfaces*, vol. 9, no. 46, pp. 41043–41054, 2017, doi: 10.1021/acsami.7b13443.
- [46] T. Hu *et al.*, “Porous structured MIL-101 synthesized with different mineralizers for adsorptive removal of oxytetracycline from aqueous solution,” *RSC Adv.*, vol. 6, no. 77, pp. 73741–73747, 2016, doi: 10.1039/c6ra11684a.
- [47] B. Huang *et al.*, “Effect of Cu(II) ions on the enhancement of tetracycline adsorption by Fe3O4@SiO2-Chitosan/graphene oxide nanocomposite,” *Carbohydr. Polym.*, vol. 157, no. Ii, pp. 576–585, 2017, doi: 10.1016/j.carbpol.2016.10.025.
- [48] N. Sumatera, R. Lubis, S. W. Saragih, and B. Wirjosentono, “zubinthinus , murr) from North Sumatera Characterization of Durian Rinds Fiber (Durio,” vol. 020069, no. December 2018, 2020.
- [49] J. Li, G. Henriksson, and G. Gellerstedt, “Lignin depolymerization/repolymerization and its critical role for delignification of aspen wood by steam explosion,” *Bioresour. Technol.*, vol. 98, no. 16, pp. 3061–3068, 2007, doi: 10.1016/j.biortech.2006.10.018.
- [50] Z. Rafiee, “Fabrication of efficient Zn-MOF/COF catalyst for the Knoevenagel condensation reaction,” *J. Iran. Chem. Soc.*, vol. 18, no. 10, pp. 2657–2664, 2021, doi: 10.1007/s13738-021-02221-z.
- [51] Z. Luo *et al.*, “Metal–Organic Framework (MOF)-based

- Nanomaterials for Biomedical Applications," *Curr. Med. Chem.*, vol. 26, no. 18, pp. 3341–3369, 2018, doi: 10.2174/0929867325666180214123500.
- [52] A. S. Eltaweil, H. M. Elshishini, Z. F. Ghatass, and G. M. Elsubrui, "Ultra-high adsorption capacity and selective removal of Congo red over aminated graphene oxide modified Mn-doped UiO-66 MOF," *Powder Technol.*, vol. 379, pp. 407–416, 2021, doi: 10.1016/j.powtec.2020.10.084.
- [53] A. Li, Y. Tong, H. Song, and X. Chen, "Compositional and Structural Evolutions of Zn-Based Metal-Organic Frameworks during Pyrolysis," *J. Phys. Chem. C*, vol. 122, no. 30, pp. 17278–17286, 2018, doi: 10.1021/acs.jpcc.8b04606.
- [54] J. Weber, N. Du, and M. D. Guiver, "Influence of intermolecular interactions on the observable porosity in intrinsically microporous polymers," *Macromolecules*, vol. 44, no. 7, pp. 1763–1767, 2011, doi: 10.1021/ma101447h.
- [55] K. V. Kumar *et al.*, "Characterization of the adsorption site energies and heterogeneous surfaces of porous materials," *J. Mater. Chem. A*, vol. 7, no. 17, pp. 10104–10137, 2019, doi: 10.1039/c9ta00287a.
- [56] Q. Guan *et al.*, "Preliminary study on shale gas microreservoir characteristics of the Lower Silurian Longmaxi Formation in the southern Sichuan Basin, China," *J. Nat. Gas Sci. Eng.*, vol. 31, pp. 382–395, 2016, doi: 10.1016/j.jngse.2016.03.023.
- [57] M. S. Hosseinirad, M. A. Ghasemzadeh, and M. S. Sharif, "Multi-component synthesis of spiro[indoline-3,4'-pyrrolo[3,4-c]pyrazoles] using Zn(BDC) metal-organic frameworks as a green and efficient catalyst," *Eurasian Chem. Commun.*, vol. 1, no. 4, pp. 344–351, 2019, doi: 10.33945/SAMI/ECC.2019.4.3.
- [58] D. Gaspar *et al.*, "Nanocrystalline cellulose applied simultaneously as the gate dielectric and the substrate in flexible field effect transistors," *Nanotechnology*, vol. 25, no. 9, 2014, doi: 10.1088/0957-4484/25/9/094008.
- [59] M. H. Jannat Abadi, S. M. M. Nouri, R. Zhiani, H. D. Heydarzadeh, and A. Motavalizadehkakhky, "Removal of tetracycline from aqueous solution using Fe-doped zeolite," *Int. J. Ind. Chem.*, vol.

- 10, no. 4, pp. 291–300, 2019, doi: 10.1007/s40090-019-0191-6.
- [60] A. Komesu, M. R. W. Maciel, and R. M. Filho, “Separation and purification technologies for lactic acid - A brief review,” *BioResources*, vol. 12, no. 3, pp. 6885–6901, 2017, doi: 10.15376/biores.12.3.6885-6901.
- [61] L. ling Yu, W. Cao, S. chuan Wu, C. Yang, and J. hua Cheng, “Removal of tetracycline from aqueous solution by MOF/graphite oxide pellets: Preparation, characteristic, adsorption performance and mechanism,” *Ecotoxicol. Environ. Saf.*, vol. 164, no. January, pp. 289–296, 2018, doi: 10.1016/j.ecoenv.2018.07.110.
- [62] Z. Wang *et al.*, “Dually organic modified bentonite with enhanced adsorption and desorption of tetracycline and ciprofloxacin,” *Sep. Purif. Technol.*, vol. 274, no. May, p. 119059, 2021, doi: 10.1016/j.seppur.2021.119059.
- [63] M. T. Islam *et al.*, “Removal of methylene blue and tetracycline from water using peanut shell derived adsorbent prepared by sulfuric acid reflux,” *J. Environ. Chem. Eng.*, vol. 7, no. 1, p. 102816, 2019, doi: 10.1016/j.jece.2018.102816.
- [64] H. Saygili and F. Güzel, “Effective removal of tetracycline from aqueous solution using activated carbon prepared from tomato (*Lycopersicon esculentum* Mill.) industrial processing waste,” *Ecotoxicol. Environ. Saf.*, vol. 131, pp. 22–29, 2016, doi: 10.1016/j.ecoenv.2016.05.001.
- [65] Y. Dai, J. Li, and D. Shan, “Adsorption of tetracycline in aqueous solution by biochar derived from waste *Auricularia auricula* dregs,” *Chemosphere*, vol. 238, p. 124432, 2020, doi: 10.1016/j.chemosphere.2019.124432.
- [66] E. E. Ghadim, F. Manouchehri, G. Soleimani, H. Hosseini, S. Kimiagar, and S. Nafisi, “Adsorption properties of tetracycline onto graphene oxide: Equilibrium, kinetic and thermodynamic studies,” *PLoS One*, vol. 8, no. 11, pp. 1–10, 2013, doi: 10.1371/journal.pone.0079254.
- [67] R. B. M. Bergamini, E. B. Azevedo, and L. R. R. de Araújo, “Heterogeneous photocatalytic degradation of reactive dyes in aqueous TiO₂ suspensions: Decolorization kinetics,” *Chem. Eng. J.*,

- vol. 149, no. 1–3, pp. 215–220, 2009, doi: 10.1016/j.cej.2008.10.019.
- [68] U. A. Edet and A. O. Ifelebuegu, “Wastewater Using Recycled Brick Waste,” *Process*, vol. 8, no. 665, pp. 1–15, 2020.
- [69] Y. Chen, F. Wang, L. Duan, H. Yang, and J. Gao, “Tetracycline adsorption onto rice husk ash, an agricultural waste: Its kinetic and thermodynamic studies,” *J. Mol. Liq.*, vol. 222, pp. 487–494, 2016, doi: 10.1016/j.molliq.2016.07.090.
- [70] M. C. Tonucci, L. V. A. Gurgel, and S. F. de Aquino, “Activated carbons from agricultural byproducts (pine tree and coconut shell), coal, and carbon nanotubes as adsorbents for removal of sulfamethoxazole from spiked aqueous solutions: Kinetic and thermodynamic studies,” *Ind. Crops Prod.*, vol. 74, pp. 111–121, 2015, doi: 10.1016/j.indcrop.2015.05.003.
- [71] F. B. Scheufele, A. N. Módenes, C. Eduardo, C. Ribeiro, and F. R. Espinoza-quiñones, “Monolayer-multilayer adsorption phenomenological model : Kinetics ,” *Chem. Eng. J.*, 2015, doi: 10.1016/j.cej.2015.09.085.
- [72] W. bo Liu *et al.*, “Efficient and selective adsorption of dye in aqueous environment employing a functional Zn(II)-based metal organic framework,” *J. Solid State Chem.*, vol. 292, no. September, p. 121740, 2020, doi: 10.1016/j.jssc.2020.121740.
- [73] H. Qiu, L. Lv, B. C. Pan, Q. J. Zhang, W. M. Zhang, and Q. X. Zhang, “Critical review in adsorption kinetic models,” *J. Zhejiang Univ. Sci. A*, vol. 10, no. 5, pp. 716–724, 2009, doi: 10.1631/jzus.A0820524.
- [74] N. Yuan, H. Cai, T. Liu, Q. Huang, and X. Zhang, “Adsorptive removal of methylene blue from aqueous solution using coal fly ash-derived mesoporous silica material,” *Adsorpt. Sci. Technol.*, vol. 37, no. 3–4, pp. 333–348, 2019, doi: 10.1177/0263617419827438.
- [75] C. Hinz, “Description of sorption data with isotherm equations,” *Geoderma*, vol. 99, no. 3–4, pp. 225–243, 2001, doi: 10.1016/S0016-7061(00)00071-9.

- [76] S. Liu, “Cooperative adsorption on solid surfaces,” *J. Colloid Interface Sci.*, vol. 450, pp. 224–238, 2015, doi: 10.1016/j.jcis.2015.03.013.
- [77] A. Ebadi, J. S. Soltan Mohammadzadeh, and A. Khudiev, “What is the correct form of BET isotherm for modeling liquid phase adsorption?,” *Adsorption*, vol. 15, no. 1, pp. 65–73, 2009, doi: 10.1007/s10450-009-9151-3.
- [78] A. Dabizha and M. Kersten, “Exothermic adsorption of chromate by goethite,” *Appl. Geochemistry*, vol. 123, no. August, p. 104785, 2020, doi: 10.1016/j.apgeochem.2020.104785.
- [79] S. Chowdhury, R. Mishra, P. Saha, and P. Kushwaha, “Adsorption thermodynamics, kinetics and isosteric heat of adsorption of malachite green onto chemically modified rice husk,” *Desalination*, vol. 265, no. 1–3, pp. 159–168, 2011, doi: 10.1016/j.desal.2010.07.047.
- [80] W. Shi, Z. Wang, F. Li, Y. Xu, and X. Chen, “Multilayer adsorption of lead (Pb) and fulvic acid by Chlorella pyrenoidosa: Mechanism and impact of environmental factors,” *Chemosphere*, vol. 329, no. January, 2023, doi: 10.1016/j.chemosphere.2023.138596.
- [81] F. Guo *et al.*, “Development of CO₂ adsorption materials from recycling spent tire char via orthogonal design: Optimal solution and thermodynamic evaluation,” *Colloids Surfaces A Physicochem. Eng. Asp.*, vol. 673, no. April, p. 131749, 2023, doi: 10.1016/j.colsurfa.2023.131749.
- [82] M. S. Imla Syafiqah and H. W. Yussof, “Kinetics, isotherms, and thermodynamic studies on the adsorption of mercury (ii) ion from aqueous solution using modified palm oil fuel ash,” *Mater. Today Proc.*, vol. 5, no. 10, pp. 21690–21697, 2018, doi: 10.1016/j.matpr.2018.07.020.
- [83] S. Shanavas, A. Salahuddin Kunju, Hema Tresa Varghese, and Yohannan Panicker, “Comparison of Langmuir and Harkins-Jura Adsorption Isotherms for the Determination of Surface Area of Solids (Shanavas 2011),” *Orient. J. Chem.*, vol. 27, no. 1, 2011.
- [84] E. Misran, O. Bani, E. M. Situmeang, and A. S. Purba, “Banana stem based activated carbon as a low-cost adsorbent for methylene

- blue removal: Isotherm, kinetics, and reusability,” *Alexandria Eng. J.*, vol. 61, no. 3, pp. 1946–1955, 2022, doi: 10.1016/j.aej.2021.07.022.
- [85] K. P. D. Iyer and A. S. Kunju, “Extension of Harkins-Jura adsorption isotherm to solute adsorption,” *Colloids and Surfaces*, vol. 63, no. 3–4, pp. 235–240, 1992, doi: 10.1016/0166-6622(92)80244-V.
- [86] E. M. Abd El-Monaem, A. M. Omer, R. E. Khalifa, and A. S. Eltaweil, “Floatable cellulose acetate beads embedded with flower-like zwitterionic binary MOF/PDA for efficient removal of tetracycline,” *J. Colloid Interface Sci.*, vol. 620, pp. 333–345, 2022, doi: 10.1016/j.jcis.2022.04.010.
- [87] B. Utami, “Deteksi Cepat Residu Tetrasiklin Melalui Startec (Smart Tetracycline Residual Kit Detection),” *J. Apl. Teknol. Pangan*, vol. 5, no. 4, pp. 151–153, 2016, doi: 10.17728/jatp.184.