

## **BAB V**

### **KESIMPULAN DAN SARAN**

#### **V.1      Kesimpulan**

Dari penelitian adsorpsi senyawa *Rhodamine B* dengan menggunakan adsorben komposit bentonit alginat, dapat disimpulkan bahwa:

1. Metode iradiasi menggunakan *microwave* dalam pembuatan adsorben komposit bentonit alginat terbukti meningkatkan kapasitas adsorpsi.
2. Metode iradiasi menggunakan *microwave* dalam pembuatan adsorben komposit bentonit alginat tidak mempengaruhi kecepatan adsorpsi.
3. Berdasarkan paramater - paramater kinetika adsorpsi menunjukkan bahwa kecepatan adsorpsi senyawa *Rhodamine B* dengan menggunakan adsorben komposit bentonit alginat mengikuti model kinetika adsorpsi *Pseudo Second Order* dengan nilai  $R^2$  berkisar antara 0,9821 – 0,9992.
4. Berdasarkan paramater - paramater adsorpsi isoterm menunjukkan bahwa adsorpsi isoterm senyawa *Rhodamine B* dengan menggunakan adsorben komposit bentonit alginat mengikuti model kinetika adsorpsi Freundlich dan Flory-Huggins dengan nilai  $R^2$  berkisar antara 0,9801 – 0,9987. Hal ini menindikasikan bahwa proses adsorpsi membentuk lapisan multilayer dipermukaan yang heterogen.

5. Berdasarkan paramater – parameter termodinamika menunjukkan bahwa adsorpsi senyawa *Rhodamine B* dengan menggunakan adsorben komposit bentonit alginat terjadi secara spontan ( $\Delta G^\circ$  berkisar -9,234 sampai dengan -14,646 kJ/mol dan  $\Delta S^\circ$  berkisar 0,0326 sampai dengan 0,2097 kJ/mol.K) dan endotermis ( $\Delta H^\circ$  berkisar 1,5251 sampai dengan 52,990 kJ/mol).
6. Metode iradiasi menggunakan *microwave* dalam pembuatan adsorben komposit bentonit alginat terbukti meningkatkan kemampuan regenerasi adsorben hingga 3 kali siklus dengan %removal mencapai 70%.

## V.2 Saran

Kami menyarankan kepada peneliti yang tertarik untuk meneliti lebih lanjut mengenai adsorpsi dalam penelitian ini dapat melakukan variasi – variasi baru terhadap variabel penelitian, seperti daya iradiasi, temperatur adsorpsi, dan pH adsorpsi sehingga ditemukan variabel – variabel yang dapat meningkatkan kemampuan adsorpsi adsorben komposit bentonit alginat.

## DAFTAR PUSTAKA

1. Li, F., Chen, Y., Huang, H., Cao, W., Li, T., “Removal of rhodamine B and Cr(VI) from aqueous solutions by a polyoxometalate adsorbent,” *Chem. Eng. Res. Des.*, 2015, 100, 192–202.
2. Madrakian, T., Afkhami, A., Mahmood-Kashani, H., Ahmadi, M., “Adsorption of some cationic and anionic dyes on magnetite nanoparticles-modified activated carbon from aqueous solutions: Equilibrium and kinetics study,” *J. Iran. Chem. Soc.*, 2013, 10, 481–489.
3. Aichour, A., Zaghouane-Boudiaf, H., “Synthesis and characterization of hybrid activated bentonite/alginate composite to improve its effective elimination of dyes stuff from wastewater,” *Appl. Water Sci.*, 2020, 10.
4. Tang, Y., Wang, Q., Zhou, B., Ma, D., Ma, Z., Zhu, L., “Synthesis of Sodium Alginate Graft Poly ( acrylic acid-co-acrylamide )/ Kaolin Composite Hydrogel and the Study on its Sorption of Rhodamine B,” 2015, 23, 467–474.
5. Khan, T.A., Dahiya, S., Ali, I., “Use of kaolinite as adsorbent: Equilibrium, dynamics and thermodynamic studies on the adsorption of Rhodamine B from aqueous solution,” *Appl. Clay Sci.*, 2012, 69, 58–66.
6. Wahyu, U., Suhendi, A., “Analysis of Rhodamin B in traditional snacks by thin layer chromatography method,” 2009, 10, 148–155.
7. Taher, T., Rohendi, D., Mohadi, R., Lesbani, A., “Thermal Activated

- of Indonesian Bentonite as A Low-Cost Adsorbent for Procion Red Removal from Aqueous Solution,” *J. Pure Appl. Chem. Res.*, 2018, 7, 79–93.
8. Martinsen, A., Skjåk-Braek, G., Smidsrød, O., “Alginate as immobilization material: I. Correlation between chemical and physical properties of alginate gel beads,” *Biotechnol. Bioeng.*, 1989, 33, 79–89.
  9. Fabryanty, R., *et al.*, “Removal of crystal violet dye by adsorption using bentonite – alginate composite,” *J. Environ. Chem. Eng.*, 2017, 5, 5677–5687.
  10. Ravi, Pandey, L.M., “Enhanced adsorption capacity of designed bentonite and alginate beads for the effective removal of methylene blue,” *Appl. Clay Sci.*, 2019, 169, 102–111.
  11. Belhouchat, N., Zaghouane-Boudiaf, H., Viseras, C., “Removal of anionic and cationic dyes from aqueous solution with activated organo-bentonite/sodium alginate encapsulated beads,” *Appl. Clay Sci.*, 2017, 135, 9–15.
  12. Warsono, H.R.S., Kurniawan, W., Hinode, H., “Utilization of modified indonesia natural bentonite for dye removal,” *ASEAN J. Chem. Eng.*, 2018, 18, 13–21.
  13. Oladipo, A.A., Gazi, M., “Enhanced removal of crystal violet by low cost alginate/acid activated bentonite composite beads: Optimization and modelling using non-linear regression technique,” *J. Water Process Eng.*, 2014, 2, 43–52.

14. Christidis, G.E., Scott, P.W., Dunham, A.C., “Acid activation and bleaching capacity of bentonites from the islands of Milos and Chios, Aegean, Greece,” *Appl. Clay Sci.*, 1997, 12, 329–347.
15. Djebri, N., Boutahala, M., Chelali, N.E., Boukhalfa, N., Zeroual, L., “Enhanced removal of cationic dye by calcium alginate/organobentonite beads: Modeling, kinetics, equilibria, thermodynamic and reusability studies,” *Int. J. Biol. Macromol.*, 2016, 92, 1277–1287.
16. Zhao, X., *et al.*, “Buoyant ALG/HA/HGMs composite adsorbents for highly efficient removal of copper from aqueous solution and contaminated kaolin soil,” *Chem. Eng. J.*, 2017, 327, 244–256.
17. Oussalah, A., Boukerroui, A., “Alginate-bentonite beads for efficient adsorption of methylene blue dye,” *Euro-Mediterranean J. Environ. Integr.*, 2020, 5.
18. Rusadi, E., Mahatmanti, F.W., Sulistyaningsih, T., “Indonesian Journal of Chemical Science Preparasi Komposit Kitosan-Bentonit sebagai Adsorben Zat Warna Methyl Orange,” *Indones. J. Chem. Sci.*, 2018, 7, 1–7, [Online]. Available: <http://journal.unnes.ac.id/sju/index.php/ijcs>.
19. Matheis F.J.D.P.Tanasale, I Wayan Sutapa, R.R.T., “Ind. J. Chem. Res. , 2014, 2, 116 - 121,” 2014, 116–121.
20. Roring, S.H., Pitoi, M.M., Abidjulu, J., “Isoterm Adsorpsi Rhodamin B Pada Arang Aktif Kayu Linggu,” *J. MIPA*, 2013, 2, 40.
21. Kurniasih, M., Riapanitra, A., Rohadi, A., “Adsorpsi Rhodamin B

- dengan Adsorben Kitosan Serbuk dan Beads Kitosan,” *Sains dan Mat.*, 2014, 2, 27–33.
22. Sahara, E., Gayatri, P.S., Suarya, P., “Adsorpsi Zat Warna Rhodamin-B dalam Larutan oleh Arang Aktif Batang Tanaman Gumitir Teraktivasi Asam Fosfat,” *Indones. E-Journal Appl. Chem.*, 2018, 6, 37–45.
  23. Langmuir, I., “THE ADSORPTION OF GASES ON PLANE SURFACES OF GLASS, MICA AND PLATINUM.,” *J. Am. Chem. Soc.*, 1918, 40, 1361–1403.
  24. Ayawei, N., Ebelegi, A.N., Wankasi, D., “Modelling and Interpretation of Adsorption Isotherms,” *J. Chem.*, 2017, 2017.
  25. Raj Somera, L., Cuazon, R., Kenneth Cruz, J., Joy Diaz, L., “Kinetics and Isotherms Studies of the Adsorption of Hg(II) onto Iron Modified Montmorillonite/Polycaprolactone Nanofiber Membrane,” *IOP Conf. Ser. Mater. Sci. Eng.*, 2019, 540.
  26. Maihendra, Fadli, A., Zultiniar, “Kinetika Adsorpsi pada Penjernahan Ion Timbal Pb<sup>2+</sup> Terlarut dalam Air Menggunakan Partikel Tricalcium Phosphate,” *Jom FTEKNIK*, 2016, 3, 1–5.
  27. Sanjaya, A.S., Agustine, R.P., “STUDI KINETIKA ADSORPSI Pb MENGGUNAKAN ARANG AKTIF DARI KULIT PISANG,” *Konversi*, 2015, 4, 17.
  28. Borhan, A., Yusup, S., Lim, J.W., Show, P.L., “Characterization and modelling studies of activated carbon produced from rubber-seed shell using KOH for CO<sub>2</sub> adsorption,” *Processes*, 2019, 7.

29. Wu, F.C., Tseng, R.L., Juang, R.S., “Initial behavior of intraparticle diffusion model used in the description of adsorption kinetics,” *Chem. Eng. J.*, 2009, 153, 1–8.
30. Fabryanty, R., Valencia, C., Edi, F., Nyoo, J., “Journal of Environmental Chemical Engineering Removal of crystal violet dye by adsorption using bentonite – alginate composite,” 2017, 5, 5677–5687.
31. Fabryanty, R., *et al.*, “Removal of crystal violet dye by adsorption using bentonite – alginate composite,” *J. Environ. Chem. Eng.*, 2017, 5, 5677–5687.
32. Mourpitchai, A., Jintakosol, T., Nitayaphat, W., “Adsorption of gold ion from a solution using montmorillonite/alginate composite,” *Mater. Today Proc.*, 2018, 5, 14786–14792.
33. Laysandra, L., *et al.*, “Highly adsorptive chitosan/saponin-bentonite composite film for removal of methyl orange and Cr(VI).,” *Environ. Sci. Pollut. Res. Int.*, 2019, 26, 5020–5037.