

LAMPIRAN A

PEMBUATAN REAGEN

Larutan buffer fosfat pH 7

Ditimbang NaH_2PO_4 50 mM sebanyak 0,4024 gram dan $\text{Na}_2\text{HPO}_4 \cdot \text{H}_2\text{O}$ 50 mM sebanyak 0,9228 gram. Lalu kedua bahan dilarutkan dengan akuades dan dituang dalam *beaker* 500 ml.

Larutan Congo-red 0,1%

Ditimbang dengan teliti 0,1 gram *congo-red* kemudian dilarutkan dengan 100 ml akuades.

Larutan asam 3,5-dinitrosalisilat (DNS)

Sebanyak 1 g DNS dilarutkan dengan 50 ml akuades dalam labu takar 100ml, ditambahkan 12,5 ml NaOH 2N, 10 ml NaSO_3 0,5%, 10 ml Fenol 2% dan ditimbang sebanyak 1 ml garam Rochelle 4% ditambahkan setelah terbentuknya kompleks warna antara DNS dan gula pereduksi hasil hidrolisis.

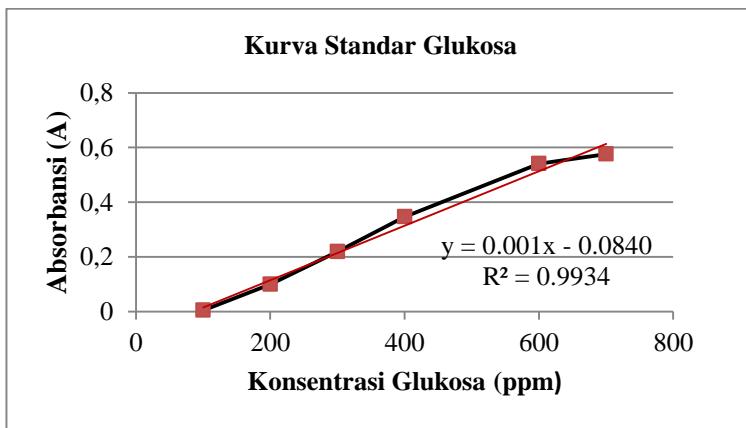
Substrat CMC 1%

Ditimbang dengan teliti sebanyak 1 gram serbuk *Carboxymethyl Cellulose* dilarutkan dalam 100 ml buffer fosfat sitrat pH 7, kemudian diautoklaf.

LAMPIRAN B
PEMBUATAN KURVA STANDAR GLUKOSA

Tabel L.2.1.Larutan Standar Glukosa

C (ppm)	A ⁵²¹
100	0,005
200	0,100
300	0,219
400	0,347
600	0,541
700	0,576



Gambar L.2.1.Kurva standar glukosa.

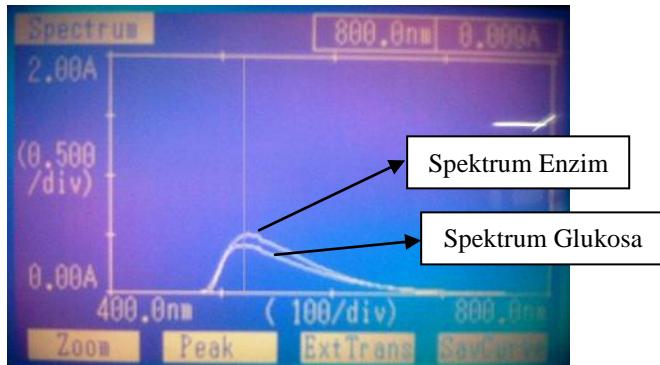
Persamaan regresi linier yang diperoleh adalah:

$$Y = 0,0010x - 0,0840$$

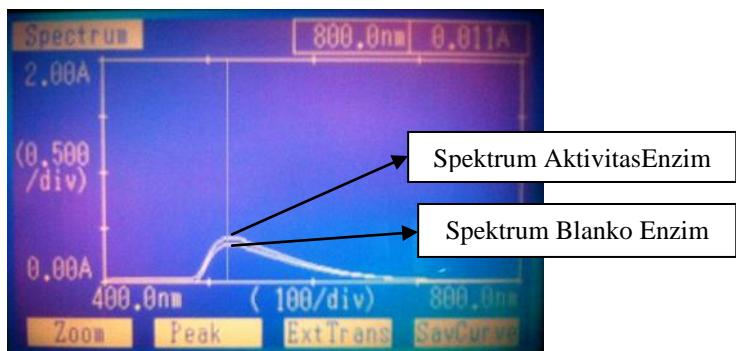
$$r = 0,9934$$

LAMPIRAN C

PENENTUAN PANJANG GELOMBANG TERPILIH



Gambar L.3.1. Hasil tumpang tindih spektrum glukosa dengan spektrum aktivitas enzim.



Gambar L.3.2. Hasil tumpang tindih spektrumblanko enzim dengan spektrum aktivitas enzim.

LAMPIRAN D
HASIL PENGUKURAN BLANKO ENZIM

Inkubasi	Rep	A ⁵²¹	Inkubasi	Rep	A ⁵²¹
24 jam	1	0,300	48 jam	1	0,325
24 jam	2	0,314	48 jam	2	0,333
24 jam	3	0,308	48 jam	3	0,342
Rata-rata		0,307	Rata-rata		0,333

Tabel L.4.1.Hasil Pengukuran Blanko Enzim

LAMPIRAN E

**PENENTUAN AKTIVITAS ENZIM SELULASE DENGAN
METODE DNS**

Persamaan regresi Linier $Y = bx + a$

Aktivitas enzim = Konsentrasi produk ($\mu\text{g/ml}$) $\times 1000 : \text{BM}_{\text{glukosa}} \times \text{Waktu inkubasi}$

Delta A = A sampel - A rata-rata blanko enzim

(Ket: A sampel diambil dari data inkubasi 24 jam replikasi 1)

Misalnya:

(Data diambil dari hasil uji aktivitas enzim inkubasi 24 jam replikasi 1)

Delta A = $0,411 - 0,037$

$$= 0,104$$

Persamaan regresi linier $Y = 0,0010x - 0,0840$

Absorbansi yang diperoleh dari delta A : 0.104

Waktu inkubasi : 30 menit

Berat molekul glukosa : 180

Konsentrasi glukosa :

$$Y = 0,0010x - 0,0840$$

Y = absorbansi

X = Konsentrasi glukosa

$$0.104 = 0,0010x - 0,0840$$

$$X = 188.322 \mu\text{g/ml} = 0,188 \text{ mg}$$

$$\text{mg glukosa} = 0,188$$

$$\text{Aktivitas Selulase} = (0,188 \times 1000) : (180 \times 30)$$

$$= 0,035 \text{ unit/ml}$$

LAMPIRAN F
DATA AKTIVITAS ENZIM SELULASE

Tabel L.6.1. Data Hasil Uji Aktivitas Ekstrak Kasar Enzim Selulase pada 24 Jam

Inkubasi	Rep	Absorbansi (A)	ΔA (A)	Cs (ppm)	mg Glukosa	AE (U/ml)
24 jam	1	0,411	0,104	188,322	0,188	0,035
24 jam	2	0,330	0,023	107,039	0,107	0,020
24 jam	3	0,361	0,054	138,147	0,138	0,026

$$\text{Rata - rata aktivitas enzim} = \frac{0,035+0,020+0,026}{3} = 0,027 \text{ unit/ml}$$

Standar Deviasi aktivitas enzim= 0,007

Tabel L.6.2. Data Hasil Uji Aktivitas Ekstrak Kasar Enzim Selulase pada 48 Jam

Inkubasi	Rep	Absorbansi (A)	ΔA (A)	Cs (ppm)	mg Glukosa	AE (U/ml)
48 jam	1	0,354	0,021	105,032	0,105	0,019
48 jam	2	0,392	0,059	143,165	0,178	0,033
48 jam	3	0,404	0,071	155,207	0,194	0,036

$$\text{Rata - rata aktivitas enzim} = \frac{0,019 + 0,033 + 0,036}{3} = 0,029 \text{ unit/ml}$$

Standar Deviasi aktivitas enzim= 0,009

LAMPIRAN G

KARAKTERISTIK FISIOLOGIS SPESIES BACILLUS

Karakteristik fisiologis spesies bakteri dari genus *Bacillus*

Table 6.9a. *Six oxidase table for Bacillus species*

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18 ^a	19	20	21	22	23	24
Gram reaction	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Chain of cells	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Motility	*	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Cell length > 3µm	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Spore position and shape	VX	VX	VX	VX	VX	VX																		
Swelling of cell body by spore	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Growth in 10% NaCl	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Anaerobic growth	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
Carbohydrates, acid from ASS:	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-d-glucose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-galactose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-mannose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-melitose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-nitrofene	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-sorbit	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-xylose	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-UNPG	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Utilization of citrate	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Urease	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Indole	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-VP	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Nitrate reduction	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Catalase hydrolysis	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Hippurate hydrolysis	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Starch hydrolysis	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	
-Oxidase	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	-	

1) <i>Bacillus antrachis</i>	9. <i>Bacillus subtilis</i>	17. <i>Bacillus laterosporus</i>
2. <i>Bacillus cereus</i> ; <i>B. cereus</i> var.	10. <i>Bacillus licheniformis</i>	18. <i>Bacillus macerans</i>
3.) <i>Bacillus mycoides</i>	11. <i>Bacillus amyloliquefaciens</i>	19. <i>Bacillus polymyxa</i>
4. <i>Bacillus thuringiensis</i>	12. <i>Bacillus coagulans</i>	20. <i>Bacillus sphaericus</i>
5. <i>Bacillus firmus</i>	13. <i>Bacillus pasteurii</i>	21. <i>Bacillus halodurans</i>
6. <i>Bacillus tentans</i>	14. <i>Bacillus brevis</i>	22. <i>Bacillus stearothermophilus</i> (Group I; Wolf & Barker, 1968; Walker & Wolf, 1971).
7. <i>Bacillus megaterium</i>	15. <i>Bacillus brevis</i>	23. <i>Bacillus stearothermophilus</i> (Group II; Wolf & Barker, 1968; Walker & Wolf, 1971).
8. <i>Bacillus pumilus</i>	16. <i>Bacillus circulans</i>	24. <i>Bacillus stearothermophilus</i> (Group III; Wolf & Barker, 1968; Walker & Wolf, 1971).

* All motile species may produce non-motile variants

T: spore terminal

V: spore central/terminal

X: spore oval/cloacoidal

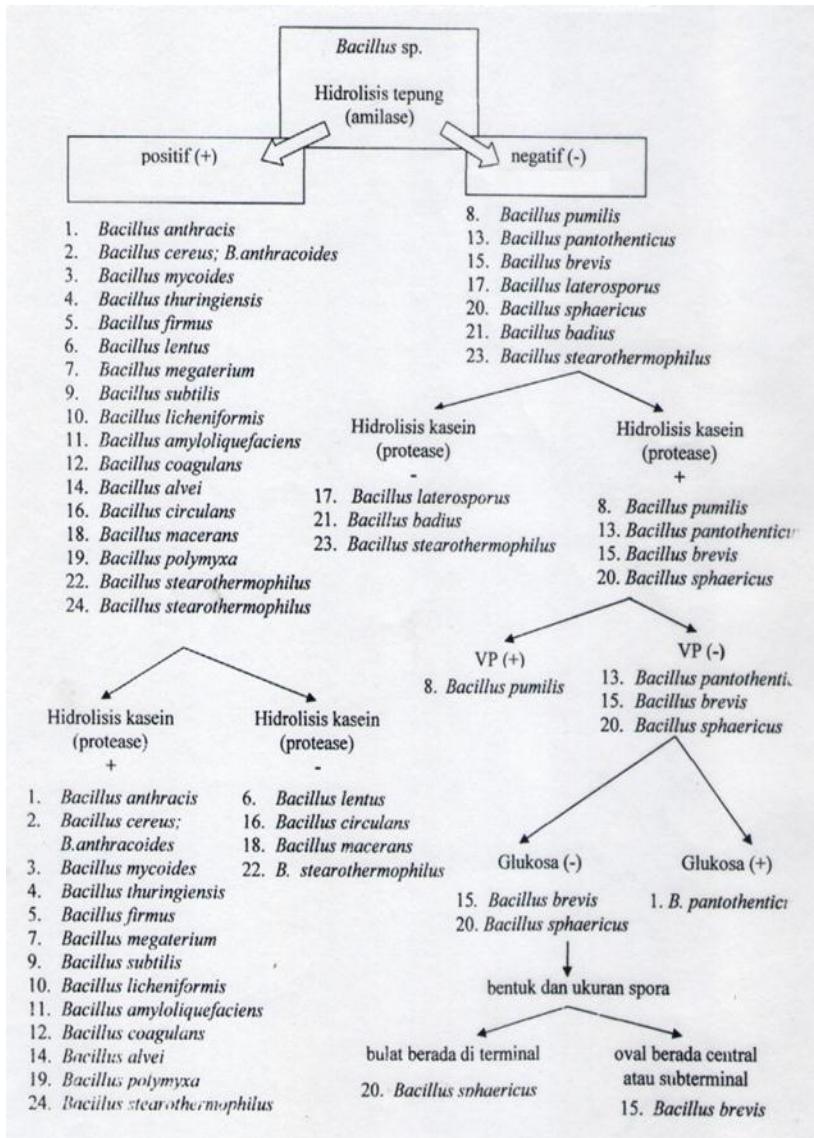
Y: spore round

Other symbols used in this table are explained in Tables S.1 and S.2 on p. 47.

Sumber: Cowan and Steel's Manual for the Identification of Medical Bacteria Third Edition (Barrow *et al.*, 1993)

LAMPIRAN H

PENGELOMPOKAN SPESIES *BACILLUS*



LAMPIRAN I

PEMBACAAN MICROBACT KIT 12A DAN 12B

Well No.	Designation	Reaction Principle	Reaction colours		Comments
			Negative	Positive	
1	Lysine	Lysine decarboxylase	Yellow	Blue-green	Green or blue is positive reaction. Bromothymol blue indicates formation of the specific amine cadaverine. Green should be regarded as a negative reaction. The pH shift indicated by bromothymol blue caused by formation of the specific amine putrescine is greater than that caused by lysine decarboxylation.
2	Ornithine	Ornithine decarboxylase	Yellow-green	Blue	H ₂ S is produced from thiosulphate. H ₂ S reacts with ferric salts in the medium to form a black precipitate.
3	H ₂ S	H ₂ S production	Straw colour	Black	Bromothymol blue indicator changes from blue to yellow when the carbohydrate is utilised to form acid.
4	Glucose	Glucose fermentation	Blue-green	Yellow	
5	Mannitol	Mannitol fermentation	Blue-green	Yellow	
6	Xylose	Xylose fermentation	Blue-green	Yellow	

Lanjutan Lampiran I

		<i>galactopyranoside</i>			
7	ONPG	ONPG by action of β -galactosidase	Colourless yellow		onitrophenol releases yellow ortho-nitrophenol.
8	Indole	Indole production from tryptophan	Colourless	Pink-red	Indole is formed from metabolism of tryptophan. Indole Kovacs reagent forms a pink-red complex with indole. Ammonium released from splitting of urea causes the pH
9	Urease	Urea hydrolysis	Straw colour	Pink-red	to rise - indicated by phenol red changing from yellow to pink-red Acetoin is produced from glucose
10	VP	Acetoin production (Voges-Proskauer reaction)	Straw colour	Pink-red	indicated by the formation of a pink-red complex after the addition of alpha-naphthol and creatine. Citrate is the sole carbon source, which if utilized results in a pH rise, indicated by bromothymol blue, with a colour change from green to blue.
11	Citrate	Citrate utilization (citrate is the only source of carbon)	Green	Blue	

Lanjutan Lampiran I

12	TDA	Production of Indolepyruvate by deamination of tryptophan	Straw colour	Cherry red	Tryptophan deaminase forms indolepyruvic acid from tryptophan which produces a brown colour in the presence of ferric ions. Indole positive organisms may produce a brown colour. This is a negative reaction.	

Well No.	Designation	Reaction Principle	Reaction colours Negative	Reaction colours Positive	Comments
1/13	Gelatin	Gelatin liquefaction	Colourless	Black	<p>Liquefaction of gelatin by proteolytic enzymes diffuses the black pigment. Solid gelatin particles which may drift across the well after rehydration should be considered as a negative reaction.</p> <p>Sodium malonate is the sole carbon source and this inhibits the conversion of succinic acid to fumaric acid.</p>

Lanjutan Lampiran I

					An organism unable to utilize this substrate results in the accumulation of succinic acid and the organism cannot grow. Bromothymol blue is the indicator. Yellow-green is indicative of a negative result.
2/14	Malonate	Malonate inhibition	Green	Blue	Utilisation of Na malonate at the same time that ammonium sulphate is utilised as the nitrogen source produces sodium hydroxide resulting in increased alkalinity and a blue colouration.
3/15	Inositol	Inositol fermentation	Blue-green	Yellow	
4/16	Sorbitol	Sorbitol fermentation	Blue-green	Yellow	
5/17	Rhamnose	Rhamnose fermentation	Blue-green	Yellow	

Lanjutan Lampiran I

6/18	Sucrose	Sucrose fermentation	Blue-green	Yellow	Bromothymol blue indicator changes from blue to yellow when the carbohydrate is fermented.
7/19	Lactose	Lactose fermentation	Blue-green	Yellow	
8/20	Arabinose	Arabinose fermentation	Blue-green	Yellow	
9/21	Adonitol	Adonitol fermentation	Blue-green	Yellow	
10/22	Raffinose	Raffinose fermentation	Blue-green	Yellow	
11/23	Salicin	Salicin fermentation	Blue-green	Yellow	
		Arginine dihydrolase			Argine dihydrolase converts arginine into ornithine, ammonia and carbon dioxide.
12/24	Arginine	24 hours	Yellow	Green-blue	This causes a pH rise as indicated by bromothymol blue. Green reactions occurring at 48 hours should be interpreted as negative
		48 hours	Yellow-green	Blue	

LAMPIRAN J

HASIL UJI FISIOLOGIS ISOLAT

Tabel 3 Isolat bakteri C pada uji amilase negatif (-)

No.	Jenis uji	C	8	13	15	17	20	21	23
1	Pewarnaan Gram	+	+	d	-	+	d	+	d
2	Bentuk sel	bt	bt	bt	bt	bt	bt	bt	bt
3	Sel yang membentuk rantai	+	+	d	-	-	+	-	-
4	Motilitas	+	+	+	+	+	+	+	+
5	Panjang sel > 3 μm	+	-	d	d	+	+	+	-
6	Bentuk dan ukuran spora	VX	VX	TYX	VX	VX	TY	VTX	VX
7	Pembentukan sel spora	-	-	+	+	+	+	-	+
Uji produksi asam dari fermentasi karbohidrat :									
8	Glucose	+	+	+	-	+	-	-	+
9	Cellubiose (Hydrolysis selulase)	-	-	-	d	-	-	-	d
10	Galactose (Gelatin)	-	+	-	-	-	-	-	-
11	Mannose (Mannitol)	+	+	d	-	d	-	-	d
12	Raffinose	+	+	d	-	d	-	-	d
13	Salicin	+	+	+	-	-	-	-	+
14	Xylose	-	-	d	-	-	-	-	-
15	ONPG (Ortho-Nitrophenyl- β -D-Galactopyranoside)	+	+	d	d	-	-	-	-
16	Citrate	+	+	-	d	-	d	-	-
17	Urea se	+	-	-	-	d	-	-	-
18	Indole	-	-	-	-	-	-	-	-
19	VP (<i>Voges Proskauer</i>)	+	+	d	-	+	-	-	+
20	Reduksi nitrat	+	-	d	d	+	d	+	-
21	Oksidase	+	-	-	-	-	+	-	-
22	Hidrolisis kasein (protein)	+	+	+	d	-	d	-	-
23	Hidrolisis lecung (amilum)	-	-	-	-	-	-	-	-
	Koefisien sebanding (%)		78	61	43	61	47	47	61

Keterangan:

- (1) spora terminal; (2) spora sentral atau subterminal; (3) spora oval.
- (8) *Bacillus pumilus*; (13) *Bacillus paniotheanicus*; (17) *Bacillus brevis*; (20) *Bacillus sphaericus*;
- (21) *Bacillus hadiae*; (23) *Bacillus stearothermophilus*

LAMPIRAN K

PERHITUNGAN ANGKA KOEFISIEN SEBANDING *BACILLUS*

Perhitungan persentase indeks kesamaan menggunakan koefisien sebanding (S_s) mencakup kesamaan positif dan negatif(Stainer *et al.*, 1986). Perhitungan menggunakan rumus:

$$S_s = \frac{a+d}{a+b+c+d} \times 100\%$$

Keterangan :

S_s = koefisien sebanding;

a = jumlah ciri positif pada kedua galur bakteri;

b = Jumlah ciri positif pada galur 1 dan negatif pada galur 2;

c = Jumlah ciri negatif pada galur 1 dan positif pada galur 2;

d = Jumlah ciri negatif pada kedua galur bakteri.

Contoh :

(koefisien sebanding dengan *Bacillus pumiluss*)

a = 15

b = 4

c = 1

d = 3

$$S_s = (15 + 3) / (15 + 4 + 1 + 3) \times 100\%$$

$$= (18 / 23) \times 100\%$$

$$= 78 \%$$