

## BAB V

### KESIMPULAN DAN SARAN

#### V.1 Kesimpulan

Dari hasil penelitian ini dapat diperoleh kesimpulan sebagai berikut:

1. Hasil karakterisasi sejumlah analisa XRD, SEM-EDX, dan FTIR menunjukkan terbentuknya MOF ZnBDC dan ZnBDC-ekstrak.
2. Kenaikan pH menyebabkan kapasitas adsorpsi MOF ZnBDC dan ZnBDC-ekstrak semakin tinggi. Kinerja adsorpsi yang paling tinggi ditunjukkan pada pH 10.
3. Semakin tinggi konsentrasi awal larutan MB, kapasitas adsorpsi MB pada MOF semakin tinggi. Penggunaan konsentrasi awal MB sebesar 1000 mg/L menunjukkan kemampuan adsorpsi MOF yang maksimal.
4. Penambahan ekstrak pada ZnBDC tidak mempengaruhi luas permukaan adsorben sehingga kenaikan kapasitas adsorpsi pada ZnBDC-ekstrak disebabkan karena interaksi kimia yang terjadi antara permukaan adsorben dengan molekul MB. Hasil studi kinetika adsorpsi yang mengikuti *model pseudo second order* memberikan konfirmasi bahwa proses adsorpsi berlangsung melalui mekanisme kimia (*chemisorption*).
5. Model isoterm adsorpsi mengikuti model Langmuir yang menunjukkan bahwa adsorpsi berlangsung secara *monolayer* pada permukaan homogen.
6. Parameter termodinamika menunjukkan bahwa proses adsorpsi bersifat spontan ( $\Delta G^0 < 0$ ), endotermik ( $\Delta H^0 > 0$ ), dan peningkatan entropi ( $\Delta S^0 > 0$ ) yang menunjukkan peningkatan suhu dapat meningkatkan kapasitas adsorpsi.

## V.2 Saran

Untuk pengembangan lebih lanjut, perlu dilakukan pengujian terhadap limbah tekstil aktual agar dapat mengevaluasi kinerja adsorben dalam kondisi lingkungan yang nyata. Selain itu, diperlukan karakterisasi lanjutan seperti *X-ray Photoelectron Spectroscopy* (XPS) untuk mengetahui komposisi kimia pada permukaan dan keadaan oksidasi unsur penyusun MOF sehingga dapat mengetahui mekanisme interaksi kimia antara gugus aktif ekstrak dengan permukaan MOF dan molekul MB. Pengujian *zeta potential* juga diperlukan untuk menentukan muatan permukaan adsorben dan stabilitas dispersi partikel dalam larutan. Selanjutnya, perlu dilakukan uji kemampuan siklus pemakaian ulang adsorben yang berguna untuk potensi keberlanjutan dan efisiensi ekonomi dalam aplikasi jangka panjang.

## DAFTAR PUSTAKA

- [1] Eka, Andro, D., Andrawan, A., Susila, U., Al Ahyar, F., Fudja Rafryanto, A., & Wahyu Indrayana, A. 2024. Limbah cangkang telur ayam sebagai adsorben zat warna methyl Blue dalam larutan. *Arramel INNOVATIVE: Journal Of Social Science Research*, 4, 3060–3069. <https://j-innovative.org/index.php/Innovative/article/view/6148>
- [2] Kementerian Negara Lingkungan Hidup. 1995. *Keputusan Menteri Negara Lingkungan Hidup Nomor: KEP-51/MENLH/10/1995 Tentang Baku Mutu Limbah Cair Bagi Kegiatan Industri*. <https://www.menlhk.go.id/>
- [3] Periyasamy, A. P. 2024. Recent advances in the remediation of textile-dye-containing wastewater: prioritizing human health and sustainable wastewater treatment. *Sustainability*, 16(2), 1–41. <https://doi.org/https://doi.org/10.3390/su16020495>
- [4] Zahro, S. F., & Adityosulindro, S. 2023. Literature review: Penggunaan bahan berbasis limbah sebagai adsorben untuk degradasi zat warna pada air limbah. *Jurnal Kesehatan Lingkungan Indonesia*, 22(3), 359–368. <https://doi.org/10.14710/jkli.22.3.359-368>
- [5] Sharma, G., AlGarni, T. S., Kumar, P. S., Bhogal, S., Kumar, A., Sharma, S., Naushad, Mu., ALOthman, Z. A., & Stadler, F. J. 2021. Utilization of Ag<sub>2</sub>O–Al<sub>2</sub>O<sub>3</sub>–ZrO<sub>2</sub> decorated onto rGO as adsorbent for the removal of congo red from aqueous solution. *Environmental Research*, 197, 1–9. <https://doi.org/https://doi.org/10.1016/j.envres.2021.111179>
- [6] Naseer, A. 2024. Role of nanocomposites and nano adsorbents for heavy metals removal and dyes. An overview. *Desalination and Water Treatment*, 320, 1–14. <https://doi.org/10.1016/j.dwt.2024.100662>
- [7] Kaur, H., Devi, N., Siwal, S. S., Alsanie, W. F., Thakur, M. K., & Thakur, V. K. 2023. Metal–organic framework-based materials for

wastewater treatment: superior adsorbent materials for the removal of hazardous pollutants. *ACS Omega*, 8(10), 9004–9030. <https://doi.org/doi:10.1021/acsomega.2c07719>

[8] Iprahim, M. S., Farag, R. S., & Tantawe, H. R. 2023. Removal of dyes from aqueous solution using zinc terephthalic acid metal-organic frameworks (Zn-TPA-MOF) based adsorbent. *Egyptian Journal of Chemistry*, 66(8), 295–305. <https://doi.org/https://doi.org/10.21608/ejchem.2022.114894.5218>

[9] Ismail, K. M., Hassan, S. S., Medany, S. S., & Hefnawy, M. A. 2024. A facile sonochemical synthesis of the Zn-based metal–organic framework for electrochemical sensing of paracetamol. *Materials Advances*, 5(14), 5870–5884. <https://doi.org/10.1039/D4MA00061G>

[10] Omana, L., Kumari K S, B., & Vijayamma, N. S. 2024. Iron Removal from Drinking Water System using Ecofriendly Synthesized Metal-Organic Frameworks. *Oriental Journal Of Chemistry*, 40(2), 528–534. <https://doi.org/10.13005/ojc/400227>

[11] Arunkumar, G., Deviga, G., Mariappan, M., Pannipara, M., Al-Sehemi, A. G., & Anthony, S. P. 2024. Natural tea extract coated porous MOF nano/microparticles for highly enhanced and selective adsorption of cationic dyes from aqueous medium. *Journal of Molecular Liquids*, 394, 1–11. <https://doi.org/https://doi.org/10.1016/j.molliq.2023.123747>

[12] Nugraheni, D. M., Kurniati, I. D., Deliara, H., & Kusuma, M. A. 2020. Kadar LDL tikus wistar setelah pemberian ekstrak kulit jeruk purut (*Citrus hystrix*). *Herb-Medicine Journal*, 3(3), 39–46. <https://doi.org/https://doi.org/10.30595/hmj.v3i3.8065>

[13] PENG, X.-J., & WU, S.-B. 2023. Incorporation of metal-organic framework MOFs-5 into the polymer monolith via the surface covalent immobilization method for enhanced capillary liquid chromatographic

separation of benzene homologues. *Chinese Journal of Analytical Chemistry*, 51(5), 1–5. <https://doi.org/10.1016/j.cjac.2022.100199>

[14] Safaei, M., Foroughi, M. M., Ebrahimpour, N., Jahani, S., Omid, A., & Khatami, M. 2019. A review on metal-organic frameworks: Synthesis and applications. *TrAC Trends in Analytical Chemistry*, 118, 401–425. <https://doi.org/10.1016/j.trac.2019.06.007>

[15] Kumar Dey, T., Hou, J., Sillanpää, M., & Kumar Pramanik, B. 2023. Metal-organic framework membrane for waterborne micro/nanoplastics treatment. *Chemical Engineering Journal*, 474, 4–21. <https://doi.org/10.1016/j.cej.2023.145715>

[16] Mohammadi, A. A., Moghanlo, S., Kazemi, M. S., Nazari, S., Ghadiri, S. K., Saleh, H. N., & Sillanpää, M. 2022. Comparative removal of hazardous cationic dyes by MOF-5 and modified graphene oxide. *Scientific Reports*, 12(1), 1–12. <https://doi.org/10.1038/s41598-022-19550-5>

[17] Song, X., Yang, P., Wu, D., Zhao, P., Zhao, X., Yang, L., & Zhou, Y. 2020. Facile synthesis of metal-organic framework UiO-66 for adsorptive removal of methylene blue from water. *Chemical Physics*, 531, 1–11. <https://doi.org/10.1016/j.chemphys.2019.110655>

[18] Eltaweil, A. S., Abd El-Monaem, E. M., Omer, A. M., Khalifa, R. E., Abd El-Latif, M. M., & El-Subruiti, G. M. 2020. Efficient removal of toxic methylene blue (MB) dye from aqueous solution using a metal-organic framework (MOF) MIL-101(Fe): isotherms, kinetics, and thermodynamic studies. *Desalination and Water Treatment*, 189, 395–407. <https://doi.org/10.5004/dwt.2020.25599>

[19] Shah, A. H., Abideen, Z. U., Maqsood, S., Rashid, F., Ullah, R., Rehman, A. U., Dildar, M., Ahmad, M., Ullah, K., Rafi, M. N., & Teng, F. 2023. Porous Cu-based metal organic framework (Cu-MOF) for highly

- selective adsorption of organic pollutants. *Journal of Solid State Chemistry*, 322, 1–9. <https://doi.org/http://dx.doi.org/10.1016/j.jssc.2023.123935>
- [20] Delpiano, G. R., Tocco, D., Medda, L., Magner, E., & Salis, A. 2021. Adsorption of malachite green and alizarin red S dyes using Fe-BTC metal organic framework as adsorbent. *International Journal of Molecular Sciences*, 22(2), 1–14. <https://doi.org/10.3390/ijms22020788>
- [21] Liu, W., Cui, G., Wang, H., Zhang, D., Wu, R., Li, L., Zhang, X., & Fan, Y. 2020. Efficient and selective adsorption of dye in aqueous environment employing a functional Zn(II)-based metal organic framework. *Journal of Solid State Chemistry*, 292, 1–9. <https://doi.org/10.1016/j.jssc.2020.121740>
- [22] Shahnawaz Khan, M., Khalid, M., & Shahid, M. 2020. What triggers dye adsorption by metal organic frameworks? The current perspectives. *Materials Advances*, 1(6), 1575–1601. <https://doi.org/https://doi.org/10.1039/D0MA00291G>
- [23] Pramudya, N. S., Hasan, R., & Junaidi, R. 2024. Kinetika pembentukan silika gel dari bottom ash sebagai adsorben. *Journal Of Social Science Research*, 4(4), 16309–16331. <https://j-innovative.org/index.php/Innovative>
- [24] Chan, Y. Y., Pang, Y. L., Lim, S., & Chong, W. C. 2021. Facile green synthesis of ZnO nanoparticles using natural-based materials: Properties, mechanism, surface modification and application. *Journal of Environmental Chemical Engineering*, 9(4), 1–27. <https://doi.org/10.1016/j.jece.2021.105417>
- [25] Diva, T. N. 2025. Advances in the synthesis and applications of MOF-5: A comprehensive review. *Journal of Chemical Reviews*, 7(1), 63–82. <https://doi.org/10.48309/JCR.2025.488450.1399>

- [26] Tang, S., Wang, Y., He, P., Wang, Y., & Wei, G. 2024. Recent advances in metal–organic framework (MOF)-based composites for organic effluent remediation. *Materials*, 17(11), 1–30. <https://doi.org/10.3390/ma17112660>
- [27] Liang, Y., Huang, H., Kou, L., Li, F., Lü, J., & Cao, H. 2018. Synthesis of metal–organic framework materials by reflux: a faster and greener pathway to achieve super-hydrophobicity and photocatalytic application. *Crystal Growth & Design*, 18(11), 1–27. <https://doi.org/10.1021/acs.cgd.8b00854>
- [28] Matuszek, K., Pankalla, E., Grymel, A., Latos, P., & Chrobok, A. 2019. Studies on the solubility of terephthalic acid in ionic liquids. *Molecules*, 25(1), 1–10. <https://doi.org/10.3390/molecules25010080>
- [29] Wu, N., Jia, T., Shi, Y.-R., Yang, Y.-J., Li, T. H., Li, F., & Wang, Z. 2020. High-performance Sn-based metal-organic frameworks anode materials synthesized by flexible and controllable methods for lithium-ion batteries. *Ionics*, 26(3), 1547–1553. <https://doi.org/10.1007/s11581-019-03392-9>
- [30] Sen, N., Shefa, N. R., Reza, K., Shawon, S. M. A. Z., & Rahman, Md. W. 2024. Adsorption of crystal violet dye from synthetic wastewater by ball-milled royal palm leaf sheath. *Scientific Reports*, 14(1), 1–17. <https://doi.org/10.1038/s41598-024-52395-8>
- [31] Shiri, M., Hosseinzadeh, M., Shiri, S., & Javanshir, S. 2024. Adsorbent based on MOF-5/cellulose aerogel composite for adsorption of organic dyes from wastewater. *Scientific Reports*, 14(1), 1–13. <https://doi.org/https://doi.org/10.1038/s41598-024-65774-y>
- [32] Rabiee, N., Atarod, M., Tavakolizadeh, M., Asgari, S., Rezaei, M., Akhavan, O., Pourjavadi, A., Jouyandeh, M., Lima, E. C., Hamed Mashhadzadeh, A., Ehsani, A., Ahmadi, S., & Saeb, M. R. 2022. Green

- metal-organic frameworks (MOFs) for biomedical applications. *Microporous and Mesoporous Materials*, 335, 1–48. <https://doi.org/10.1016/j.micromeso.2021.111670>
- [33] Svensson Grape, E., Chacón-García, A. J., Rojas, S., Pérez, Y., Jaworski, A., Nero, M., Åhlén, M., Martínez-Ahumada, E., Galetsa Feindt, A. E., Pepillo, M., Narongin-Fujikawa, M., Ibarra, I. A., Cheung, O., Baresel, C., Willhammar, T., Horcajada, P., & Inge, A. K. 2023. Removal of pharmaceutical pollutants from effluent by a plant-based metal–organic framework. *Nature Water*, 1(5), 433–442. <https://doi.org/https://doi.org/10.1038/s44221-023-00070-z>
- [34] Mahreni, M., Ramadhan, R. R., Pramadhana, M. F., Permatasari, A. P., Kurniawati, D., & Kusuma, H. S. 2022. Synthesis of metal organic framework (MOF) based Ca-alginate for adsorption of malachite green dye. *Polymer Bulletin*, 79(12), 11301–11315. <https://doi.org/https://doi.org/10.1007/s00289-022-04086-5>
- [35] Zamzamiyah, I. N., Sumeru, D., Jurusan, A., Pertanian, B., Pertanian, F., Brawijaya, U., Veteran, J., & Timur, J. 2020. Eksplorasi dan karakterisasi tanaman jeruk purut (*Citrus hystrix*) di kabupaten tulungagung. *Jurnal Produksi Tanaman*, 8(11), 1041–1049.
- [36] Adlini, M. N., & Umaroh, H. K. 2021. Karakterisasi tanaman jeruk (*Citrus sp.*) di kecamatan nibung hangus kabupaten batu bara sumatera utara. *KLOROFIL: Jurnal Ilmu Biologi Dan Terapan*, 4(1), 1–10. <https://doi.org/10.30821/kfl:jibt.v4i1.8921>
- [37] Rahmiati, N., Sari, R., & Wahyuni, T. S. 2023. Phytochemical and antioxidant activity evaluation of lime (*Citrus aurantifolia*) juice powder. *Jurnal Farmasi Galenika (Galenika Journal of Pharmacy) (e-Journal)*, 9(2), 197–207. <https://doi.org/10.22487/j24428744.2023.v9.i2.16347>

- [38] Septiyan, T., Gofarana Wilar, & Nasrul Wathoni. 2021. Herbal medication of recurrent aphthous stomatitis: a narrative review. *Bioscientia Medicina : Journal of Biomedicine and Translational Research*, 6(1), 1318–1323. <https://doi.org/https://doi.org/10.37275/bsm.v6i1.438>
- [39] Abdullah, A., Haron, N., Yusof, H., Camalxaman, S. N., Mohamed, E., & Rambely, A. S. 2023. Antioxidant potential and phytochemical content from peels of three different citrus species: A mini review. *Journal of Sustainability Science and Management*, 18(12), 123–135. <https://doi.org/http://dx.doi.org/10.46754/jssm.2023.12.011>
- [40] Rahayu, M. S., Mellaratna, W. P., & Najah, N. 2024. The preliminary study on the antifungal effect of Kaffir lime (*Citrus hystrix* DC) peel extract against *Malassezia furfur*. *International Journal of Secondary Metabolite*, 11(3), 486–493. <https://doi.org/https://doi.org/10.21448/ijsm.1395562>
- [41] Mohideen, M., Mahadi, N. N. S. J., Suhaimi, N. A. N., Kamaruzaman, N. A., & Azlan, A. Y. H. N. 2022. Antibacterial properties of essential oil extracted from kaffir lime (*Citrus hystrix*) peel. *Biomedical and Pharmacology Journal*, 15(1), 179–186. <https://doi.org/https://dx.doi.org/10.13005/bpj/2353>
- [42] Wijaya, Y. A., Widyadinata, D., Irawaty, W., & Ayucitra, A. 2017. Fraction of phenolic compounds from kaffir lime (*Citrus hystrix*) peel extract and evaluation of antioxidant activity. *Reaktor*, 17(3), 111–117. <https://doi.org/10.14710/reaktor.17.3.111-117>
- [43] Assefa, H., Singh, S., Olu, F. E., Dhanjal, D. S., Mani, D., Khan, N. A., Singh, J., & Ramamurthy, P. C. 2024. Advances in adsorption technologies for hexavalent chromium removal: Mechanisms, materials, and optimization strategies. *Desalination and Water Treatment*, 319, 1–12. <https://doi.org/https://doi.org/10.1016/j.dwt.2024.100576>

- [44] de Carvalho Costa, L. R., Jurado-Davila, I. V., Oliveira, J. T. De, Nunes, K. G. P., Estumano, D. C., de Oliveira, R. A., Carissimi, E., & F  ris, L. A. 202. Exploring key parameters in adsorption for effective fluoride removal: A comprehensive review and engineering implications. *Applied Sciences* (Switzerland), 14(5), 1–22. <https://doi.org/https://doi.org/10.3390/app14052161>
- [45] Haleem, A., Shafiq, A., Chen, S.-Q., & Nazar, M. 2023. A comprehensive review on adsorption, photocatalytic and chemical degradation of dyes and nitro-compounds over different kinds of porous and composite materials. *Molecules*, 28(3), 1–38. <https://doi.org/10.3390/molecules28031081>
- [46] Au, V. K.-M. 2020. Recent Advances in the Use of Metal-Organic Frameworks for Dye Adsorption. *Frontiers in Chemistry*, 8, 1–7. <https://doi.org/https://doi.org/10.3389/fchem.2020.00708>
- [47] Li, X., Han, D., Zhang, M., Li, B., Wang, Z., Gong, Z., Liu, P., Zhang, Y., & Yang, X. 2019. Removal of toxic dyes from aqueous solution using new activated carbon materials developed from oil sludge waste. *Colloids and Surfaces A: Physicochemical and Engineering Aspects*, 578, 1–10. <https://doi.org/http://dx.doi.org/10.1016/j.colsurfa.2019.05.066>
- [48] Sonmez, G., & Akyuz, L. 2024. In situ preparation and characterization of Cr-MOF-alginates for methylene blue through the adsorption process. *Journal of Water Process Engineering*, 62, 1–44. <https://doi.org/10.1016/j.jwpe.2024.105318>
- [49] Shafti, D. M., Dahlan, I., & Din, A. T. M. 2024. A review of the effectiveness of metal–organic frameworks in removing dye effluents. *Water Practice & Technology*, 19(12), 4699–4733. <https://doi.org/10.2166/wpt.2024.279>

- [50] Ho, Y. S., & McKay, G. 1998. Sorption of dye from aqueous solution by peat. *Chemical Engineering Journal*, 70(2), 115–124. [https://doi.org/10.1016/S0923-0467\(98\)00076-1](https://doi.org/10.1016/S0923-0467(98)00076-1)
- [51] Guo, X., & Wang, J. 2019. Comparison of linearization methods for modeling the Langmuir adsorption isotherm. *Journal of Molecular Liquids*, 296, 1–36. <https://doi.org/10.1016/j.molliq.2019.111850>
- [52] Alfonso Herrera, L. Á., Camarillo Reyes, P. K., Huerta Flores, A. M., Martínez, L. T., & Rivera Villanueva, J. M. 2020. BDC-Zn MOF sensitization by MO/MB adsorption for photocatalytic hydrogen evolution under solar light. *Materials Science in Semiconductor Processing*, 109, 1–9. <https://doi.org/10.1016/j.mssp.2020.104950>
- [53] Wijaya, R., Andersan, G., Permatasari Santoso, S., & Irawaty, W. 2020. Green reduction of graphene oxide using kaffir lime peel extract (*Citrus hystrix*) and its application as adsorbent for methylene blue. *Scientific Reports*, 10(1), 1–9. <https://doi.org/10.1038/s41598-020-57433-9>
- [54] Elsherbiny, A. S., Rady, A., Abdelhameed, R. M., & Gemeay, A. H. 2023. Efficiency and selectivity of cost-effective Zn-MOF for dye removal, kinetic and thermodynamic approach. *Environmental Science and Pollution Research*, 30(49), 106860–106875. <https://doi.org/10.1007/s11356-023-25919-4>
- [55] Stuart, B. H. 2004. *Infrared spectroscopy: Fundamentals and applications*. Wiley & Sons. <https://doi.org/10.1002/0470011149>
- [56] Purba, C. C., Mayangsari, Y., Setyaningsih, W., Chansuwan, W., & Sirinupong, N. 2024. Bioactive compounds of *Citrus hystrix* peel ethanolic extract and their antioxidant potential under hydrogen peroxide-induced oxidative stress in Caco-2 cells. *Future Foods*, 9, 1–11. <https://doi.org/10.1016/j.fufo.2024.100350>

- [57] Irawaty, W., & Ayucitra, A. 2018. Assessment on antioxidant and in vitro antidiabetes activities of different fractions of *Citrus hystrix* peel. *International Food Research Journal*, 25(6), 2467–2477.
- [58] Hwang, J., Yan, R., Oschatz, M., & Schmidt, B. V. K. J. 2018. Solvent mediated morphology control of zinc MOFs as carbon templates for application in supercapacitors. *Journal of Materials Chemistry A*, 6(46), 23521–23530. <https://doi.org/10.1039/C8TA07700B>
- [59] Niu, C., Zhang, N., Hu, C., Zhang, C., Zhang, H., & Xing, Y. 2021. Preparation of a novel citric acid-crosslinked Zn-MOF/chitosan composite and application in adsorption of chromium(VI) and methyl orange from aqueous solution. *Carbohydrate Polymers*, 258, 1–10. <https://doi.org/10.1016/j.carbpol.2021.117644>
- [60] Lubinska-Szczygeł, M., Kuczyńska-Łażewska, A., Rutkowska, M., Polkowska, Ż., Katrich, E., & Gorinstein, S. 2023. Determination of the major by-products of *Citrus hystrix* peel and their characteristics in the context of utilization in the industry. *Molecules*, 28(6), 2596. <https://doi.org/10.3390/molecules28062596>
- [61] Ojo, M. A. 2022. Tannins in foods: Nutritional implications and processing effects of hydrothermal techniques on underutilized hard-to-cook legume seeds—a review. *Preventive Nutrition and Food Science*, 27(1), 14–19. <https://doi.org/10.3746/pnf.2022.27.1.14>
- [62] Wang, N., Li, S., Li, Z., Gong, Y., & Li, X. 2023. A Zn(II)–metal–organic framework sased on 4-(4-carboxy phenoxy) phthalate acid as luminescent sensor for detection of acetone and tetracycline. *Molecules*, 28(3), 1–14. <https://doi.org/10.3390/molecules28030999>
- [63] Healy, C., Patil, K. M., Wilson, B. H., Hermanspahn, L., Harvey-Reid, N. C., Howard, B. I., Kleinjan, C., Kolien, J., Payet, F., Telfer, S. G., Kruger, P. E., & Bennett, T. D. 2020. The thermal stability of metal-organic

- frameworks. *Coordination Chemistry Reviews*, 419, 1–20. <https://doi.org/10.1016/j.ccr.2020.213388>
- [64] More, M. S., Bodkhe, G. A., Singh, F., Kim, M., & Shirsat, M. D. 2023. Metal–organic framework-reduced graphene oxide (Zn-BDC@rGO) composite for selective discrimination among ammonia, carbon monoxide, and sulfur dioxide. *Applied Physics A*, 129(12), 1–11. <https://doi.org/10.1007/s00339-023-07103-0>
- [65] de Souza, D. A., da Silva, P. H. de A., da Silva, F. P., Romaguera-Barcelay, Y., Ferreira, R. D., Araujo Junior, E. A., Nascimento, J. F. de L., da Costa, F. F., Takeno, L. L., Leyet Ruiz, Y., Dominguez, L. A., Brito, W. R., & Nobre, F. X. 2024. Easy and fast obtention of ZnO by thermal decomposition of zinc acetate and Its photocatalytic properties over Rhodamine B dye. *Colorants*, 3(3), 229–252. <https://doi.org/10.3390/colorants3030016>
- [66] Aunan, E., Affolter, C. W., Olsbye, U., & Lillerud, K. P. 2021. Modulation of the thermochemical stability and adsorptive properties of MOF-808 by the selection of non-structural ligands. *Chemistry of Materials*, 33(4), 1471–1476. <https://doi.org/10.1021/acs.chemmater.0c04823>
- [67] Adams, F. 200. X-Ray absorption and diffraction | overview. In *Encyclopedia of Analytical Science* (pp. 365–377). Elsevier. <https://doi.org/10.1016/B0-12-369397-7/00668-3>
- [68] Mansoor Al-Jarsha, H. Y., & Ghareeb, M. M. 2024. Application of taguchi orthogonal array in optimization of the synthesis and crystallinity of metal organic framework 5 (MOF 5). *Tropical Journal of Pharmaceutical Research*, 22(12), 2473–2482. <https://doi.org/10.4314/tjpr.v22i12.8>
- [69] Stanciu, M.-C., Tanasă, F., & Teacă, C.-A. 2025. Crystallinity changes in modified cellulose substrates evidenced by spectral and x-ray

- diffraction data. *Polysaccharides*, 6(2), 1–32.  
<https://doi.org/10.3390/polysaccharides6020030>
- [70] Rápó, E., & Tonk, S. 2021. Factors affecting synthetic dye adsorption; desorption studies: a review of results from the last five years (2017–2021). *Molecules*, 26(17), 1–31.  
<https://doi.org/10.3390/molecules26175419>
- [71] Tejada-Tovar, C., Villabona-Ortíz, Á., & Gonzalez-Delgado, Á. D. 2021. Adsorption of zzo-anionic dyes in a solution using modified coconut (*Cocos nucifera*) mesocarp: kinetic and equilibrium study. *Water*, 13(10), 1–20. <https://doi.org/10.3390/w13101382>
- [72] Budiarti, L. Y., Yasmina, A., Nurikwan, P. W., Prayudi, M. O. S., Firisa, M. R., & Kangsudarmanto, Y. 2022. Antibacterial activity of infused peel of kaffir lime, manurun banana, and pineapple against the number of *Staphylococcus aureus* and *Escherichia coli* colonies. *IOP Conference Series: Earth and Environmental Science*, 976(1), 1–9.  
<https://doi.org/10.1088/1755-1315/976/1/012034>
- [73] Khan, M. F., & Khan, M. A. 2023. Plant-derived metal nanoparticles (PDMNPs): synthesis, characterization, and oxidative stress-mediated therapeutic actions. *Future Pharmacology*, 3(1), 252–295.  
<https://doi.org/10.3390/futurepharmacol3010018>
- [74] Villagrán, Z., Anaya-Esparza, L. M., Velázquez-Carriles, C. A., Silva-Jara, J. M., Ruvalcaba-Gómez, J. M., Aurora-Vigo, E. F., Rodríguez-Lafitte, E., Rodríguez-Barajas, N., Balderas-León, I., & Martínez-Esquivias, F. 2024. Plant-based extracts as reducing, capping, and stabilizing agents for the green synthesis of inorganic nanoparticles. *Resources*, 13(6), 1–24.  
<https://doi.org/10.3390/resources13060070>
- [75] Al-Maliky, E. A., Gzar, H. A., & Al-Azawy, M. G. 2021. Determination of point of zero charge (PZC) of concrete particles

- adsorbents. *IOP Conference Series: Materials Science and Engineering*, 1184(1), 1–8. <https://doi.org/10.1088/1757-899X/1184/1/012004>
- [76] K., A. K., Bisoi, A., M., Y., Shobham, Jujaru, M., Panwar, J., & Gupta, S. 2024. Unveiling the dye adsorption capability of *Moringa oleifera* functionalized hybrid porous MOF–GO composites: *in vitro* and *in silico* ecotoxicity assessment *via* antibacterial and molecular docking studies. *Environmental Science: Water Research & Technology*, 10(8), 1938–1963. <https://doi.org/10.1039/D4EW00185K>
- [77] Tatarchuk, T., Soltys, L., & Macyk, W. 2023. Magnetic adsorbents for removal of pharmaceuticals: A review of adsorption properties. *Journal of Molecular Liquids*, 384, 1–22. <https://doi.org/10.1016/j.molliq.2023.122174>
- [78] Kosmulski, M. 2023. The pH dependent surface charging and points of zero charge. X. update. *Advances in Colloid and Interface Science*, 319, 1–26. <https://doi.org/10.1016/j.cis.2023.102973>
- [79] Patel, K. 2023. Structural organization of peptides. In *De Novo Peptide Design* (pp. 1–33). Elsevier. <https://doi.org/10.1016/B978-0-323-99917-5.00008-1>
- [80] Calinsky, R., & Levy, Y. 2024. Aromatic residues in proteins: Re-evaluating the geometry and energetics of  $\pi$ – $\pi$ , cation– $\pi$ , and CH– $\pi$  interactions. *The Journal of Physical Chemistry B*, 128(36), 8687–8700. <https://doi.org/10.1021/acs.jpcc.4c04774>
- [81] Qiu, B., Shao, Q., Shi, J., Yang, C., & Chu, H. 2022. Application of biochar for the adsorption of organic pollutants from wastewater: Modification strategies, mechanisms and challenges. *Separation and Purification Technology*, 300, 1–9. <https://doi.org/10.1016/j.seppur.2022.121925>

- [82] Osman, H. H., Rodríguez-Hernández, P., Muñoz, A., & Manjón, F. J. 2025. A unified theory of electron-rich and electron-deficient multicenter bonds in molecules and solids: a change of paradigms. *Journal of Materials Chemistry C*, 13(8), 3774–3803. <https://doi.org/10.1039/D4TC04441J>
- [83] Aaddouz, M., Azzaoui, K., Akartasse, N., Mejdoubi, E., Hammouti, B., Taleb, M., Sabbahi, R., & Alshahateet, S. F. 2023. Removal of methylene blue from aqueous solution by adsorption onto hydroxyapatite nanoparticles. *Journal of Molecular Structure*, 1288, 1–23. <https://doi.org/10.1016/j.molstruc.2023.135807>
- [84] AttahDaniel, E. B., Mtunzi, F. M., Wankasi, D., Ayawei, N., Dikio, E. D., & Diagboya, P. N. 2022. Relative empirical evaluation of the aqueous sequestration of methylene blue using benzene-1,4-dicarboxylic acid-linked lanthanum and zinc metal organic frameworks. *Water, Air, & Soil Pollution*, 233(11), 1–16. <https://doi.org/10.1007/s11270-022-05912-2>
- [85] Bagherzadeh, M., Salehi, G., & Rabiee, N. 2024. Rapid and efficient removal of methylene blue dye from aqueous solutions using extract-modified Zn–Al LDH. *Chemosphere*, 350, 1–10. <https://doi.org/10.1016/j.chemosphere.2023.141011>
- [86] Wang, L., Shi, C., Wang, L., Pan, L., Zhang, X., & Zou, J.-J. 2020. Rational design, synthesis, adsorption principles and applications of metal oxide adsorbents: a review. *Nanoscale*, 12(8), 4790–4815. <https://doi.org/10.1039/C9NR09274A>
- [87] Saleh, T. A. 2022. *Kinetic models and thermodynamics of adsorption processes: classification*. 34, 65–97. <https://doi.org/10.1016/B978-0-12-849876-7.00003-8>
- [88] Agbor Tabi, G., Ngouateu Rene Blaise, L., Daouda, K., Naphtali Odogu, A., Aime Victoire, A., Nsami Julius, N., & Joseph Mbadcam, K. 2022. Non-linear modelling of the adsorption of indigo carmine dye from

- wastewater onto characterized activated carbon/volcanic ash composite. *Arabian Journal of Chemistry*, 15(1), 1–12. <https://doi.org/10.1016/j.arabjc.2021.103515>
- [89] Murphy, O. P., Vashishtha, M., Palanisamy, P., & Kumar, K. V. 2023. A review on the adsorption isotherms and design calculations for the optimization of adsorbent mass and contact time. *ACS Omega*, 8(20), 17407–17430. <https://doi.org/10.1021/acsomega.2c08155>
- [90] Pereira, S. K., Kini, S., Prabhu, B., & Jeppu, G. P. 2023. A simplified modeling procedure for adsorption at varying pH conditions using the modified Langmuir–Freundlich isotherm. *Applied Water Science*, 13(1), 1–13. <https://doi.org/10.1007/s13201-022-01800-6>
- [91] Butyrskaya, E. 2024. Understanding the mechanism of monolayer adsorption from isotherm. *Adsorption*, 30(6), 1395–1406. <https://doi.org/10.1007/s10450-024-00512-4>
- [92] Rath, B. S., & Kumar, P. S. 2021. Application of adsorption process for effective removal of emerging contaminants from water and wastewater. *Environmental Pollution*, 280, 1–19. <https://doi.org/10.1016/j.envpol.2021.116995>
- [93] Üzek, R., Şenel, S., & Denizli, A. 2022. Investigation of thermodynamic, kinetic, and isothermal parameters for the selective adsorption of bisphenol A. *ACS Omega*, 7(22), 18940–18952. <https://doi.org/10.1021/acsomega.2c01975>
- [94] Alharbi, K. N., Alsulami, R. A., Alhakami, M. H., Baata, M., Alotaibi, M. F., Alshehri, S., Alzahrani, S. M., Alwafi, A. M., Aljufareen, M. A., Albarqi, M. M., Alnajem, K. N., & Almouqli, M. S. 2025. Investigation of adsorption isotherms and thermodynamic models of uranium biosorption from aqueous solutions by *Rumex Acetosella*. *Journal*

*of Radioanalytical and Nuclear Chemistry*, 334(3), 2251–2270.  
<https://doi.org/10.1007/s10967-025-10031-3>

[95] Perry, Robert. H. 2008. *Perry's Chemical Engineers' Handbook 8th Edition*.