

BAB V KESIMPULAN

1. Rasio massa adsorbent:volume limbah cair sintesis MG yang menghasilkan persen removal tertinggi adalah 1:100 gram/mL.
2. Suhu adsorpsi yang menghasilkan persen removal tertinggi adalah 60°C.
3. Pada saat awal (0-20 menit) persen removal MG mengalami kenaikan secara drastis, selanjutnya kenaikan tersebut semakin lama semakin kecil hingga konstan pada saat kesetimbangan (120 menit).
4. Persamaan kinetika adsorpsi MG menggunakan TBA dari kulit kayu bakau mengikuti persamaan orde 2 semu sebagai berikut:

$$\text{Linier: Orde 2 semu: } \frac{t}{q_t} = 0,1605 \cdot t + 1,3076$$

$$\text{Non-linier: Orde 2 semu: } qt = \frac{0,8124 \cdot t}{1 + 0,1312 \cdot t}$$

5. Persamaan isotherm adsorpsi MG menggunakan TBA dari kulit kayu bakau mengikuti persamaan Langmuir sebagai berikut:

$$q_e = 312,5 \frac{0,0154 \cdot C_e}{1 + 0,0154 \cdot C_e}$$

6. Adsorpsi MG menggunakan adsorbent TBA dari kulit kayu bakau merupakan adsorpsi kimia (*chemisorption*).

DAFTAR PUSTAKA

1. Sirvastava, S., Ranggana, S., Roy, D., *Toxicological Effect of MG Aquat, Toxicol*, 2004, 15: p.219-238.
2. Sukardjo, S., *Some aspect of mangrove ecology*, 1978, Training Materials for Forestry Officer, Integreted Development of the Sundarbans Reserved Forest, Rome, FAO.
3. Hamidah, S., *Rendemen dan Kadar Tanin Kulit Kayu Bakau. (Rhizophora mucronata Lamck) dari Daerah Takisung*, Jurnal Hutan Tropis Borneo, 2006, 18: p.15-23.
4. J. Zhang, Y. Li, C. Zhang, Y. Jing, *Adsorption of Malachite Green from Aqueous Solution onto Carbon Prepared from Arundo Donax Root*, J. Hazard. Mater, 2008, 150: p.774–782.
5. K.V.K. Rao, *Inhibition of DNA Synthesis in Primary Rat Hepatocyte Cultures by Malachite Green: A New Liver Tumor Promoter*, Toxicol Letters, 1995, 81: p.107-113.
6. F.J. Foster, L. Woodbury, *The Use of Malachite Green as a Fish Fungicide and Antiseptic*, Prog. Fish-Cult, 2011, 18: p.7-9.
7. https://en.wikipedia.org/wiki/Malachite_green#/media/File:Malachite_green_structure.svg diakses pada 30 Mei 2019.
8. S.J. Culp, L.R. Blankenship, D.F. Kusewitt, D.R. Doerge, L.T. Mulligan, F.A. Beland, *Toxicity and metabolism of MG and leucoMG during short-term feeding to Fischer 344 rats and B6C3F1 mice*, Chem. Biol. Interact., 1999, 122: p.153–170.
9. V.K. Gupta, A. Mittal, L. Krishnan, V. Gajbe, *Adsorption kinetics and column operations for the removal and recovery of MG from wastewater using bottom ash*, Sep. Purif. Technol., 2004, 40: p. 87–96.
10. Emrah Bulut, Mahmut Özacara, İ. Ayhan Şengil, *Adsorption of Malachite Green Onto Bentonite: Equilibrium and Kinetic Studies and Process Design*, Microporous and Mesoporous Materials, 2008, 115: p.234-246.
11. Tandy, E., *Kemampuan Adsorbent Limbah Lateks Karet Alam Terhadap Minyak Pelumas Dalam Air*. Jurnal Teknik Kimia USU, 2012. 1: p. 2.
12. Saragih, S.A., *Pembuatan dan Karekterisasi Karbon Aktif dari Batubara Riau sebagai Adsorbent*, 2008, Jakarta: Universitas Indonesia.
13. A Alberty, R.A., 1987, *Physical Chemistry*, 7th, John willey and Sons.

14. Kratochvil, D., Volesky, B., *Advances in the biosorption of heavy metals*, Trends Biotechnol, 1998, 16: p. 291-300.
15. Mueller-Harvey, I., *Unravelling the conundrum of tannins in animal nutrition and health*. J. Sci. Food Agric, 2006, 86: p.2010–2037.
16. Haslam and Edwin, *Vegetable tannins-lessons of a phytochemical lifetime*. Phytochemistry, 2007, 68(22-24): p. 2713-2721.
17. Gouveia, L. and A.C. Oliveira, *Microalgae as a raw material for biofuels production*. J Ind Microbiol Biotechnol, 2009, 36: p. 269–274.
18. Carter, F.L., A.M. Carlo, and J.B. Stanley, *Termiticidal Components of Wood Extracts Methyljuglone from Diospyros Virginia*. Agriculture, 1978, 26(4): p. 869-873.
19. Ismarani, *Potensi senyawa tannin dalam menunjang produksi ramah lingkungan*. Jurnal Agribisnis dan Pengembangan Wilayah, 2012, 3 (2): p. 46-55.
20. Wisnubroto, D.S., *Pengolahan Logam berat dari limbah cair dengan Tannin*, 2002, Banten: Pusat Pengolahan dan Pengembangan Limbah Radioaktif.
21. Browning, B.L., *Methods of Wood Chemistry*. Vol. I, II. 1966, New York: Interscience Publishers.
22. https://upload.wikimedia.org/wikipedia/commons/9/96/Tannic_acid.png, diakses pada 30 Mei 2019
23. Ahadi, M.R., *Kandungan Tanin Terkondensasi dan Laju Dekomposisi pada Serasah Daun Rhizospora mucronata lamk pada Ekosistem Tambak Tumpangsari, Purwakarta, Jawa Barat*. Skripsi, 2003.
24. Agus Haerudin, Farida F. F., *Limbah Serutan Kayu Matoa (Pometia Pinnata) Sebagai Zat Warna Alam Pada Kain Batik Serat Selulosa*, Indonesian Ministry of Industry, 2017.
25. Priscilla C. Veggi, Julian Martinez, and M. Angela A. Meireles., 2010, *Fundamental of Microwave Extraction*.
26. Be´atrice Kaufmann and Philippe Christen, *Recent Extraction Techniques for Natural Products: Microwave-assisted Extraction and Pressurised Solvent Extraction*. Phytochem. Anal., 2002, 13: p. 105–113
27. Bernabas IJ, Dean JR, Fowlis IA and Owen SP. 1995.*Extraction of Polycyclic Aromatic Hydrocarbon From Highly Contaminated Soils Using Microwave Energy*. Analyst 120 : p. 1897-1904

28. Luque-Garcia, J.L. and M.D. Luque de Castro, 2003. *Where is Microwave Based Analytical Treatment For Solid Sample Pre-Treatment Going?*. Trends Anal. Chem., 22: p. 90-99.
29. P. Tatke and Y. Jaiswal, *An Overview of Microwave Assisted Extraction and its Applications in Herbal Drug Research*, 2011, 5 (1): p. 21-31
30. Ismadji, S., Sunarso, J., Febrianto, J., Kosasih, A.N., Yi-Hsu, J., Indraswati, N., *Equilibrium and kinetic studies in adsorption of heavy metals using biosorbent : A summary of recent studies*. Journal of Hazardous Materials, 2009, 162: p. 616-645.
31. Estiaty, L.M., *Keseimbangan dan kinetika adsorpsi ion Cu²⁺ pada zeolit-H*. Riset, Geologi dan Pertambangan, 2012, 22 (2): p. 127-141.
32. Rozaini C. A., Jain K., Oo C. W. Tan K. W., Tan L. S, Azraa A., Tong K. S., *Optimization of Nickel and Copper Ions Removal by Modified Mangrove Barks*, International Journal of Chemical Engineering and Applications, 2010, 1.
33. J. Sánchez-Martín, M. González-Velasco, J. Beltrán-Heredia, J. Gragera-Carvajal, J. Salguero-Fernández, *Novel Tannin-Based Adsorbent in Removing Cationic Dye (Methylene Blue) from Aqueous Solution*, Journal of Hazardous Materials, 2010, 174: p. 9-16.
34. C.W. Ooa, M.J. Kassim, A. Pizzi, *Characterization and Performance of Rhizophora Apiculata Mangrove Polyflavonoid Tannins in The Adsorption of Copper (II) And Lead (II)*, Industrial Crops and Products, 2009, 30: p. 152-161.
35. Tan Lean Seey, Mohd Jain Noordin Mohd Kassim, *Acidic and Basic Dyes Removal by Adsorption on Chemically Treated Mangrove Barks*, International Journal of Applied Science and Technology, 2012, 2.
36. Sofía Arellano-Cárdenas, Socorro López-Cortez, Maribel Cornejo-Mazón, Juan Carlos Mares-Gutiérrez, *Study of Malachite Green Adsorption by Organically Modified Clay Using A Batch Method*, Applied Surface Science, 2013, 280, 1: p. 74-78.
37. S.D. Khattri, M.K. Singh, *Removal of MG from dye wastewater using neem sawdust by adsorption*, Journal of Hazardous Materials, 2009, p. 1089–1094.
38. B.H. Hameed dan M.I. El-Khaiary, *Malachite Green Adsorption by Rattan Sawdust: Isotherm, Kinetic and Mechanism Modeling*, Journal of Hazardous Materials, 2008, 159: p. 574-579.

39. Rais Ahmad, Rajeev Kumar, *Adsorption Studies of Hazardous Malachite Green Onto Treated Ginger Waste*, Journal of Environmental Management, 2010, 91: p. 1032-1038.
40. Y.Önal, C.Akmil-Başar, Ç.Sarıcı-Özdemir, *Investigation kinetics mechanisms of adsorption malachite green onto activated carbon*, Journal of Hazardous Materials, 2007, 146: p. 194-203.
41. Pantoja-Castroa Mayra A., González-Rodríguez Horacio, *Study by Infrared Spectroscopy and Thermogravimetric Analysis of Tannins and Tannic Acid*, Revista latinoamericana de química, 2012, p. 109–110.
42. Aibin Zhang, Jiongjiong Li, Shifeng Zhang, Youbing Mu, Wei Zhang and Jianzhang Li, *Characterization and acid-catalysed depolymerization of condensed tannins derived from larch bark*, RSC Advances, 2017, 7: p. 35135–35146.
43. Y. M. Li, X. Miao, Z. G. Wei, J. Cui, S. Y. Li, R. M. Han, Y. Zhang, W. Wei, *Iron-Tannic Acid Nanocomplexes: Facile Synthesis and Application for Removal Of Methylene Blue from Aqueous Solution*, Digest Journal of Nanomaterials and Biostructures, 2016, 11(4): p. 1045-1061.
44. Ruina Zhang, Lin Li, Junxin Liu, *Synthesis and characterization of ferric tannate as a novel porous adsorptive-catalyst for nitrogen removal from wastewater*, RSC Advances, 2015, 51.
45. V.K. Garg, Rakesh Kumar, Renuka Gupta, *Removal of Malachite Green Dye from Aqueous Solution by Adsorption Using Agro-Industry Waste: A Case Study of Prosopis Cineraria*, Dyes and Pigments, 2004, 62: p. 1-10.
46. Zenon Pawlak, *Surface Tribochemistry and Activated Processes*, Tribology and Interface Engineering Series, 2003, 45: p. 161-215.