

BAB V

KESIMPULAN

1. Pelarut etanol yang menghasilkan kadar produk tanin yang paling tinggi adalah etanol 75% dengan kadar tanin sebesar 1,63% .
2. Waktu dan jumlah siklus ekstraksi yang menghasilkan kadar produk tanin paling tinggi adalah pada waktu 60 detik dan jumlah siklus ekstraksi 5 kali dengan kadar tanin sebesar 1,63%.
3. Rasio massa adsorben:volume limbah (b/v) yang memberikan persen penurunan kadar ion Cr(VI) paling tinggi adalah 0,03 gram:25 mL.
4. Persamaan kinetika adsorpsi yang cocok untuk adsorpsi ion Cr(VI) dalam limbah sintetis menggunakan *Tannin Based Adsorbent* (TBA) dari daun jambu biji adalah kinetika adsorpsi orde 2 semu dengan persamaan:

$$\frac{d_{qt}}{d_t} = 3,775 \cdot (50,25 - q_t)^2 \text{ dengan } R^2 \text{ sebesar } 0,9557.$$

DAFTAR PUSTAKA

1. Volesky B. Detoxification of metal-bearing effluents: biosorption for the next century. *Hydrometallurgy* 2001; 59: 203-216
2. Robinson, T. *Kandungan Organik Tumbuhan Tinggi*. 1995. Penerjemah: Kosasih Padmawinata. Bandung:ITB.
3. Ekholm, P., Virkki, L., Ylinen, M., Johanson, L. *The effect of Phytic Acid and Some Natural Chelating Agents of Solubility of Mineral Elements in Oat Bran*. 2003.
4. Chemat F, Albert-Vian M, Zill-e-Huma Y-J (2009) Microwave assisted separation: green chemistry in action. In: Pearlman JT (ed) Green chemistry research trends. *Nova Science Publishers*, New York, pp 33-62
5. Hasrianti, "Adsorpsi Ion Cd²⁺ dan Cr⁶⁺ pada Limbah Cair menggunakan Kulit Singkong" 2012, Program Pasca Sarjana, Universitas Hasanuddin, Makassar.
6. Hartati, I., Riwayati, I., dan Kurniasari, L."Potensi Xanthate Pulpa Kopi sebagai Adsorben pada Pemisahan Ion Timbal dari Limbah Industri Batik", *Jurnal Momentum* 2011, Vol. 7, No. 2.
7. Bielicka A, I Bojanowska, A Wisniewski. 2005. Two faces of chromium-pollutant and bioelement. *Journal of Environmental Studies*. 14 (1): 5-10.
8. Matsuo, T., Hanarnure, N., Shimol, K., Nakamura, Y., and Tomita, I. 1993. Identification of (+) gallic acid as a-bio antimutagenic compound in *Psidium guajava* leaves. *Phytochemistry* 36: 1027-1029
9. Giner-Chavez, B.I. dan Cannas, A. *Tannins: Chemical Structural The Struktur od Hydrolysable Tannins*. [cited 26 November 2015. Available from : <http://www.anSci.cornell.edu/plant/toxicagents/> tannin/ image/int.blg.gif.corner university]. 2001.

10. Jain, T., Jain, V., Pandey, R., Vyas, A., & Shukla, S. S. (2009). Microwave Assisted Extraction for Phytoconstituents – An Overview. *Asian Journal Research Chemistry*, 1 (2), 19-25.
11. Guenther, E., 1952. “The essential Oils Volume 5” D. Van Nostrand Company Inc. New York. 420 pp.
12. Ganzler, K., Salgo’ , A., & Valko’ , K ., 1986, Microwave extraction: A novel sample preparation method for chromatography. *Journal of Chromatography*, 371, 299-306.
13. Salas, P. G., Aranzazu, M.-S., Antonio, S.-C., & Alberto, F.-G. (2010, December 3). Phenolic-Compound-Extraction Systems for Fruit and Vegetable Samples. *Molecules* , 15, pp. 8813-8826.
14. Mc. Cabe and Smith, “Unit Operation of Chemical Engineering”, McGraw Hill Book Company, 5th edition 1993, 810, Singapore.
15. Alberty, R.A., 1987, Physical Chemistry, 7th, John wiley and Sons.
16. Z.-x. Jin, B.-Q. Wang and Z.-J. Chen, *Medicinal Plants Research* 2010, 4, 2229-2234.
17. N. Rhazi, H. Hannache, M. Oumam, A. Sesbou, B. Charrier, A. Pizzi and F. C.-E. Bouhtoury, *Arabian Journal of Chemistry* 2015, 1878-5352.
18. Kunnambath, P.M., Thirumalaisamy, S., *Characterization and utilization of tannins extract for the selective adsorption of Ni (II)* Journal of Chemistry, 2015. **2015**: p. 1-9.
19. Suzuki, Y., Sawada, K., Chihara, K., *Adsorption characteristics of tannin for heavy metal ions*. 2004, Meiji University: Kawasaki, Japan.
20. A. Nakajima and Y. Baba. 2004. *Mechanism of hexavalent chromium adsorption by persimmon tannin gel*. Japan. Water Research.

21. A. Sengil and M. Ozacar, *Journal of Hazardous Materials* 2008, 277-285.
22. Markom, M., Hasan, M., Daud, W.R.W., Singh, H., and Jaim, J.M., (2007), Extraction of hydrolysable tannins from *Phyllanthus niruri* Linn:Effects of solvents and extraction methods, *Separation and Purification Technology*, 52, pp. 487-496.
23. Rohmawai, L., "Studi Kinetika Adsorpsi Merkuri (II) Pada Biomassa Daun Enceng Gondok (*Eichhornia crassipes*)" Skripsi 2008, Jurusan Kimia, Fakultas Sains dan Teknologi, Universitas Islam Negeri Malang, Malang.
24. Yu Lin H, KuoYH, Lin YL, Chiang W. Antioxidative effect and active component from leaves of lotus (*Nelumbo nucifera*). *Journal of Agricultural and Food Chemistry* 2009; 57 : 6623-6629.
25. Mokhtarpour A, Naserian AA, Tahmasbi AM, Valizadeh R. Extraction of Phenolic Compounds and Tannins from Pistachio By-products. *Animal Science* 2014: 1330-1338.
26. Baldosano HY, Castillo Ma BMG, Elloran CDH, Bacani FT. Effect of Particle Size, Solvent and Extraction Time on Tannin Extract from *Spondias purpurea* Bark Through Soxhlet Extraction. 2015. Vol 3.
27. P. C. Veggi, J. Martinez and M. A. A. Meireles, 2013, "Fundamental of Microwave Extraction", Brazil.
28. Pantoja-Castroa, M.A., González-Rodríguez,H., *Study By Infrared Spectroscopyand Thermogravimetric Analysis of Tannins and Tannic Acid*. Latinoamer.Quím, 2011. **39**: p. 107-112.
29. Argun ME, Dursun S, Ozdemir C, Karatas M. Heavy metal adsorption by modified oak sawdust: Thermodynamics and kinetics. *Journal of Hazardous Material* 2007: 77-85.

30. Mu'jizah, S., "Pembuatan dan Karakterisasi Karbon Aktif dari Biji Kelor (*Moringa oleifera. Lamk*) dengan NaCl sebagai Bahan Pengaktif" Skripsi 2010, Jurusan Kimia, Fakultas Sains dan Teknologi, Universitas Islam Negeri, Malang.